

Diethylmalonic acid, 2,3-dichlorophenyl tetradecyl ester

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|----------------------|---|
| Inchi: | InChI=1S/C27H42Cl2O4/c1-4-7-8-9-10-11-12-13-14-15-16-17-21-32-25(30)27(5-2,6-3)26 |
| InchiKey: | DBKGAHBVGXUAIE-UHFFFAOYSA-N |
| Formula: | C27H42Cl2O4 |
| SMILES: | CCCCCCCCCCCCCOC(=O)C(CC)(CC)C(=O)Oc1cccc(Cl)c1Cl |
| Mol. weight [g/mol]: | 501.53 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -219.25 | kJ/mol | Joback Method |
| hf | -916.85 | kJ/mol | Joback Method |
| hfus | 65.50 | kJ/mol | Joback Method |
| hvap | 105.08 | kJ/mol | Joback Method |
| log10ws | -9.73 | | Crippen Method |
| logp | 8.950 | | Crippen Method |
| mcvol | 406.890 | ml/mol | McGowan Method |
| pc | 829.55 | kPa | Joback Method |
| rinpol | 3241.00 | | NIST Webbook |
| rinpol | 3241.00 | | NIST Webbook |
| tb | 1078.01 | K | Joback Method |
| tc | 1323.54 | K | Joback Method |
| tf | 652.09 | K | Joback Method |
| vc | 1.575 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 1355.04 | J/molxK | 1078.01 | Joback Method |
| cpg | 1418.34 | J/molxK | 1282.61 | Joback Method |
| cpg | 1408.37 | J/molxK | 1241.69 | Joback Method |
| cpg | 1397.16 | J/molxK | 1200.77 | Joback Method |
| cpg | 1384.60 | J/molxK | 1159.85 | Joback Method |
| cpg | 1370.60 | J/molxK | 1118.93 | Joback Method |
| cpg | 1427.16 | J/molxK | 1323.54 | Joback Method |
| dvisc | 0.0000103 | Paxs | 1078.01 | Joback Method |

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|-------|-----------|------|---------|---------------|
| dvisc | 0.0000135 | Paxs | 1007.02 | Joback Method |
| dvisc | 0.0000184 | Paxs | 936.04 | Joback Method |
| dvisc | 0.0000263 | Paxs | 865.05 | Joback Method |
| dvisc | 0.0000402 | Paxs | 794.06 | Joback Method |
| dvisc | 0.0000668 | Paxs | 723.08 | Joback Method |
| dvisc | 0.0001238 | Paxs | 652.09 | Joback Method |

Sources

| | |
|------------------------|---|
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U370042&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307l |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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