

# Succinic acid, 2,3-dichlorophenyl 1-phenylpropyl ester

Inchi:	InChI=1S/C19H18Cl2O4/c1-2-15(13-7-4-3-5-8-13)24-17(22)11-12-18(23)25-16-10-6-9-14
InchiKey:	SVFWHCSUURTOEV-UHFFFAOYSA-N
Formula:	C19H18Cl2O4
SMILES:	CCC(OC(=O)CCC(=O)Oc1cccc(Cl)c1Cl)c1ccccc1
Mol. weight [g/mol]:	381.25

## Physical Properties

Property code	Value	Unit	Source
gf	-179.48	kJ/mol	Joback Method
hf	-511.73	kJ/mol	Joback Method
hfus	42.71	kJ/mol	Joback Method
hvap	90.46	kJ/mol	Joback Method
log10ws	-6.18		Crippen Method
logp	5.373		Crippen Method
mvol	270.410	ml/mol	McGowan Method
pc	1759.49	kPa	Joback Method
rinpol	2677.00		NIST Webbook
rinpol	2677.00		NIST Webbook
tb	924.44	K	Joback Method
tc	1160.38	K	Joback Method
tf	570.93	K	Joback Method
vc	1.024	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	767.94	J/molxK	924.44	Joback Method
cpg	779.26	J/molxK	963.76	Joback Method
cpg	789.28	J/molxK	1003.09	Joback Method
cpg	798.06	J/molxK	1042.41	Joback Method
cpg	805.63	J/molxK	1081.73	Joback Method
cpg	812.02	J/molxK	1121.06	Joback Method
cpg	817.29	J/molxK	1160.38	Joback Method
dvisc	0.0003766	Paxs	570.93	Joback Method

dvisc	0.0002237	Paxs	629.85	Joback Method
dvisc	0.0001452	Paxs	688.77	Joback Method
dvisc	0.0001009	Paxs	747.68	Joback Method
dvisc	0.0000740	Paxs	806.60	Joback Method
dvisc	0.0000565	Paxs	865.52	Joback Method
dvisc	0.0000447	Paxs	924.44	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U389934&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U389934&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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