

# 2-Methyl-1,5-naphthalene diisocyanate

**Inchi:** InChI=1S/C13H8N2O2/c1-9-5-6-10-11(13(9)15-8-17)3-2-4-12(10)14-7-16/h2-6H,1H3  
**InchiKey:** WWWOJJNVCOVBNW-UHFFFAOYSA-N  
**Formula:** C13H8N2O2  
**SMILES:** Cc1ccc2c(N=C=O)cccc2c1N=C=O  
**Mol. weight [g/mol]:** 224.21  
**CAS:** 56775-58-7

## Physical Properties

Property code	Value	Unit	Source
chs	-6125.40 ± 4.20	kJ/mol	NIST Webbook
hf	70.72	kJ/mol	Joback Method
hfs	-133.00 ± 4.20	kJ/mol	NIST Webbook
hvap	69.50	kJ/mol	Joback Method
log10ws	-12.72		Crippen Method
logp	3.083		Crippen Method
mcvol	165.310	ml/mol	McGowan Method
pc	3131.49	kPa	Joback Method
tb	690.78	K	Joback Method
tc	929.33	K	Joback Method

## Sources

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>  
**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)  
**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)  
**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>  
**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C56775587&Units=SI>

## Legend

**chs:** Standard solid enthalpy of combustion

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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