

# Sebacic acid, undecyl 1-phenyl-2,2,2-trifluoromethylethyl ester

**Inchi:** InChI=1S/C29H45F3O4/c1-2-3-4-5-6-7-10-13-19-24-35-26(33)22-17-11-8-9-12-18-23-27  
**InchiKey:** YJLBGUSDJCRPFW-UHFFFAOYSA-N  
**Formula:** C29H45F3O4  
**SMILES:** CCCCCCCCCCOC(=O)CCCCCCCCC(=O)OC(c1ccccc1)C(F)(F)F  
**Mol. weight [g/mol]:** 514.66

## Physical Properties

Property code	Value	Unit	Source
gf	-746.16	kJ/mol	Joback Method
hf	-1497.32	kJ/mol	Joback Method
hfus	68.78	kJ/mol	Joback Method
hvap	96.60	kJ/mol	Joback Method
log10ws	-9.91		Crippen Method
logp	9.028		Crippen Method
mvol	415.900	ml/mol	McGowan Method
pc	735.62	kPa	Joback Method
rinpol	2964.00		NIST Webbook
rinpol	2964.00		NIST Webbook
tb	1036.32	K	Joback Method
tc	1282.35	K	Joback Method
tf	576.52	K	Joback Method
vc	1.637	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1453.49	J/molxK	1036.32	Joback Method
cpg	1472.40	J/molxK	1077.33	Joback Method
cpg	1489.56	J/molxK	1118.33	Joback Method
cpg	1505.13	J/molxK	1159.34	Joback Method
cpg	1519.22	J/molxK	1200.34	Joback Method
cpg	1531.98	J/molxK	1241.35	Joback Method
cpg	1543.53	J/molxK	1282.35	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U416808&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U416808&amp;Units=SI</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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