

Pentafluoropropanamide, N-hexyl-

Inchi:	InChI=1S/C9H14F5NO/c1-2-3-4-5-6-15-7(16)8(10,11)9(12,13)14/h2-6H2,1H3,(H,15,16)
InchiKey:	WMVFFBSJXLBQIO-UHFFFAOYSA-N
Formula:	C9H14F5NO
SMILES:	CCCCCNC(=O)C(F)(F)C(F)(F)F
Mol. weight [g/mol]:	247.21

Physical Properties

Property code	Value	Unit	Source
gf	-983.00	kJ/mol	Joback Method
hf	-1286.25	kJ/mol	Joback Method
hfus	26.34	kJ/mol	Joback Method
hvap	42.13	kJ/mol	Joback Method
log10ws	-3.53		Crippen Method
logp	2.880		Crippen Method
mcvol	158.070	ml/mol	McGowan Method
pc	2079.33	kPa	Joback Method
rinpola	1084.00		NIST Webbook
rinpola	1084.00		NIST Webbook
tb	499.25	K	Joback Method
tc	656.56	K	Joback Method
tf	301.57	K	Joback Method
vc	0.648	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	394.12	J/mol×K	499.25	Joback Method
cpg	406.85	J/mol×K	525.47	Joback Method
cpg	418.90	J/mol×K	551.69	Joback Method
cpg	430.29	J/mol×K	577.91	Joback Method
cpg	441.06	J/mol×K	604.12	Joback Method
cpg	451.23	J/mol×K	630.34	Joback Method
cpg	460.84	J/mol×K	656.56	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U407335&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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