

# Benzenesulfonamide, N-(phenylsulfonyl)-

<b>Other names:</b>	Dibenzenesulfonamide Benzenesulfonic acid imide Dibenzenesulfimide Dibenzolsulfimid Dibenzolsulfimide Dibenzenesulfonimide N-(phenylsulphonyl)benzenesulphonamide
<b>Inchi:</b>	InChI=1S/C12H11NO4S2/c14-18(15,11-7-3-1-4-8-11)13-19(16,17)12-9-5-2-6-10-12/h1-11
<b>InchiKey:</b>	OVQABVAKPIYHIG-UHFFFAOYSA-N
<b>Formula:</b>	C12H11NO4S2
<b>SMILES:</b>	O=S(=O)(NS(=O)(=O)c1ccccc1)c1ccccc1
<b>Mol. weight [g/mol]:</b>	297.35
<b>CAS:</b>	2618-96-4

## Physical Properties

Property code	Value	Unit	Source
gf	-572.71	kJ/mol	Joback Method
hf	-671.18	kJ/mol	Joback Method
hfus	42.77	kJ/mol	Joback Method
hvap	90.56	kJ/mol	Joback Method
log10ws	-2.84		Crippen Method
logp	1.354		Crippen Method
mcvol	198.580	ml/mol	McGowan Method
pc	5213.15	kPa	Joback Method
tb	673.05	K	Joback Method
tc	899.23	K	Joback Method
tf	407.62	K	Joback Method
vc	0.778	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	494.12	J/molxK	673.05	Joback Method
cpg	508.55	J/molxK	710.75	Joback Method

cpg	521.65	J/mol×K	748.44	Joback Method
cpg	533.45	J/mol×K	786.14	Joback Method
cpg	543.96	J/mol×K	823.83	Joback Method
cpg	553.21	J/mol×K	861.53	Joback Method
cpg	561.23	J/mol×K	899.23	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2618964&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2618964&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/125-251-4/Benzenesulfonamide-N-phenylsulfonyl.pdf>

Generated by Cheméo on 2024-04-20 02:37:14.375833408 +0000 UTC m=+15869883.296410722.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.