

Urea, (2-hydroxyethyl)-

Other names:	1-Ethanolurea («beta»-Hydroxyethyl)urea N-(«beta»-Hydroxyethyl)urea N-(2-Hydroxyethyl)urea (2-Hydroxyethyl)urea Monoethanolurea Monoethylolurea
Inchi:	InChI=1S/C3H8N2O2/c4-3(7)5-1-2-6/h6H,1-2H2,(H3,4,5,7)
InchiKey:	CLAHOZSYMRNIPY-UHFFFAOYSA-N
Formula:	C3H8N2O2
SMILES:	NC(=O)NCCO
Mol. weight [g/mol]:	104.11
CAS:	2078-71-9

Physical Properties

Property code	Value	Unit	Source
gf	-135.52	kJ/mol	Joback Method
hf	-282.80	kJ/mol	Joback Method
hfus	19.51	kJ/mol	Joback Method
hvap	62.77	kJ/mol	Joback Method
log10ws	0.28		Crippen Method
logp	-1.353		Crippen Method
mcvol	80.530	ml/mol	McGowan Method
pc	6093.99	kPa	Joback Method
tb	536.79	K	Joback Method
tc	726.30	K	Joback Method
tf	370.24	K	Joback Method
vc	0.292	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	187.65	J/mol×K	536.79	Joback Method
cpg	194.19	J/mol×K	568.37	Joback Method

cpg	200.41	J/mol×K	599.96	Joback Method
cpg	206.31	J/mol×K	631.54	Joback Method
cpg	211.90	J/mol×K	663.13	Joback Method
cpg	217.18	J/mol×K	694.71	Joback Method
cpg	222.18	J/mol×K	726.30	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C2078719&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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