

# 2-chloro-2,3,3-trimethylpentane

<b>Inchi:</b>	InChI=1S/C8H17Cl/c1-6-7(2,3)8(4,5)9/h6H2,1-5H3
<b>InchiKey:</b>	LKNHZRZTEOZKEA-UHFFFAOYSA-N
<b>Formula:</b>	C8H17Cl
<b>SMILES:</b>	CCC(C)(C)C(C)(C)Cl
<b>Mol. weight [g/mol]:</b>	148.67
<b>CAS:</b>	69078-89-3

## Physical Properties

Property code	Value	Unit	Source
gf	10.23	kJ/mol	Joback Method
hf	-241.69	kJ/mol	Joback Method
hfus	5.84	kJ/mol	Joback Method
hvap	35.20	kJ/mol	Joback Method
log10ws	-3.19		Crippen Method
logp	3.440		Crippen Method
mcvol	135.820	ml/mol	McGowan Method
pc	2537.93	kPa	Joback Method
tb	413.41	K	Joback Method
tc	607.51	K	Joback Method
tf	283.75 ± 0.20	K	NIST Webbook
vc	0.510	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	266.40	J/mol×K	413.41	Joback Method
cpg	282.16	J/mol×K	445.76	Joback Method
cpg	296.97	J/mol×K	478.11	Joback Method
cpg	310.86	J/mol×K	510.46	Joback Method
cpg	323.88	J/mol×K	542.81	Joback Method
cpg	336.09	J/mol×K	575.16	Joback Method
cpg	347.54	J/mol×K	607.51	Joback Method
dvisc	0.0149900	Paxs	214.68	Joback Method
dvisc	0.0051495	Paxs	247.80	Joback Method

dvisc	0.0022759	Paxs	280.92	Joback Method
dvisc	0.0011949	Paxs	314.04	Joback Method
dvisc	0.0007095	Paxs	347.17	Joback Method
dvisc	0.0004613	Paxs	380.29	Joback Method
dvisc	0.0003213	Paxs	413.41	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.chemic.org/files/research/kdb/mol/mol1646.mol">https://www.chemic.org/files/research/kdb/mol/mol1646.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C69078893&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C69078893&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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