

# 1-N-NONYLNAPHTHALENE

<b>Inchi:</b>	InChI=1S/C19H36/c1-2-3-4-5-6-7-8-12-17-14-11-15-18-13-9-10-16-19(17)18/h17-19H,2-
<b>InchiKey:</b>	JYZIPVEKDVGHQ-OTWHNJEPSA-N
<b>Formula:</b>	C19H36
<b>SMILES:</b>	CCCCCCCCC1CCCC2CCCC12
<b>Mol. weight [g/mol]:</b>	254.41
<b>CAS:</b>	26438-26-6

## Physical Properties

Property code	Value	Unit	Source
gf	174.49	kJ/mol	Joback Method
hf	-334.87	kJ/mol	Joback Method
hfus	33.91	kJ/mol	Joback Method
hvap	58.09	kJ/mol	Joback Method
log10ws	-6.84		Crippen Method
logp	6.734		Crippen Method
mvol	256.850	ml/mol	McGowan Method
nfpaf	%!d(float64=2)		KDB
pc	1356.63	kPa	Joback Method
tb	660.01	K	Joback Method
tc	854.43	K	Joback Method
tf	321.45	K	Joback Method
vc	0.981	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	770.43	J/molxK	660.01	Joback Method
cpg	795.54	J/molxK	692.41	Joback Method
cpg	819.27	J/molxK	724.82	Joback Method
cpg	841.68	J/molxK	757.22	Joback Method
cpg	862.81	J/molxK	789.63	Joback Method
cpg	882.72	J/molxK	822.03	Joback Method
cpg	901.47	J/molxK	854.43	Joback Method
dvisc	0.0038419	Paxs	321.45	Joback Method

dvisc	0.0017656	Paxs	377.88	Joback Method
dvisc	0.0009931	Paxs	434.30	Joback Method
dvisc	0.0006376	Paxs	490.73	Joback Method
dvisc	0.0004486	Paxs	547.16	Joback Method
dvisc	0.0003370	Paxs	603.58	Joback Method
dvisc	0.0002659	Paxs	660.01	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.chemie.org/research/kdb/hcprop/showprop.php?cmpid=812">https://www.chemie.org/research/kdb/hcprop/showprop.php?cmpid=812</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemie.com/doc/models/crippen_log10ws">https://www.chemie.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>nf<sub>paf</sub>:</b>	NFPA Fire Rating
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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