

# 2(3H)-Furanone, 3-bromodihydro-

<b>Other names:</b>	«alpha»-Bromo-«gamma»-butyrolactone 2-Bromo-4-butyrolactone 2-Bromo-«gamma»-butyrolactone «alpha»-Bromobutyric acid, «gamma»-lactone «alpha»-Bromobutyrolactone 2-Bromo-4-butanolide 2-Bromobutyrolactone 3-Bromodihydro-2-furanone 3-Bromodihydro-2(3H)-furanone Butyric acid, 2-bromo-4-hydroxy-, gamma-lactone 3-Bromo-2-oxotetrahydrofuran NSC 56959
<b>Inchi:</b>	InChI=1S/C4H5BrO2/c5-3-1-2-7-4(3)6/h3H,1-2H2
<b>InchiKey:</b>	LFJJGHGXHXXDFT-UHFFFAOYSA-N
<b>Formula:</b>	C4H5BrO2
<b>SMILES:</b>	O=C1OCCC1Br
<b>Mol. weight [g/mol]:</b>	164.99
<b>CAS:</b>	5061-21-2

## Physical Properties

Property code	Value	Unit	Source
gf	-175.04	kJ/mol	Joback Method
hf	-308.78	kJ/mol	Joback Method
hfus	12.82	kJ/mol	Joback Method
hvap	39.95	kJ/mol	Joback Method
log10ws	-0.80		Crippen Method
logp	0.697		Crippen Method
mcvol	81.300	ml/mol	McGowan Method
pc	5678.82	kPa	Joback Method
tb	467.13	K	Joback Method
tc	710.18	K	Joback Method
tf	300.33	K	Joback Method
vc	0.290	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	148.75	J/mol×K	467.13	Joback Method
cpg	158.59	J/mol×K	507.64	Joback Method
cpg	167.92	J/mol×K	548.15	Joback Method
cpg	176.75	J/mol×K	588.65	Joback Method
cpg	185.07	J/mol×K	629.16	Joback Method
cpg	192.87	J/mol×K	669.67	Joback Method
cpg	200.16	J/mol×K	710.18	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	411.20	K	0.80	NIST Webbook
tbrp	369.50 ± 0.50	K	0.70	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5061212&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5061212&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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