

# Cyclohexanol, 5-methyl-2-propyl, trans, cis

**Inchi:** InChI=1S/C10H20O/c1-3-4-9-6-5-8(2)7-10(9)11/h8-11H,3-7H2,1-2H3/t8-,9-,10+/m1/s1  
**InchiKey:** VWXNPISBYOISDJ-BBBLOLIVSA-N  
**Formula:** C10H20O  
**SMILES:** CCCC1CCC(C)CC1O  
**Mol. weight [g/mol]:** 156.27

## Physical Properties

Property code	Value	Unit	Source
gf	-94.47	kJ/mol	Joback Method
hf	-388.32	kJ/mol	Joback Method
hfus	19.72	kJ/mol	Joback Method
hvap	54.34	kJ/mol	Joback Method
log10ws	-2.80		Crippen Method
logp	2.584		Crippen Method
mcvol	146.770	ml/mol	McGowan Method
pc	2637.96	kPa	Joback Method
rinpol	1187.00		NIST Webbook
rinpol	1187.00		NIST Webbook
tb	530.59	K	Joback Method
tc	716.30	K	Joback Method
tf	262.18	K	Joback Method
vc	0.545	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	370.12	J/mol×K	530.59	Joback Method
cpg	447.90	J/mol×K	685.35	Joback Method
cpg	433.83	J/mol×K	654.39	Joback Method
cpg	419.04	J/mol×K	623.44	Joback Method
cpg	403.49	J/mol×K	592.49	Joback Method
cpg	387.19	J/mol×K	561.54	Joback Method
cpg	461.25	J/mol×K	716.30	Joback Method
dvisc	0.0001533	Paxs	530.59	Joback Method

dvisc	0.0002407	Paxs	485.85	Joback Method
dvisc	0.0004142	Paxs	441.12	Joback Method
dvisc	0.0008057	Paxs	396.38	Joback Method
dvisc	0.0018562	Paxs	351.65	Joback Method
dvisc	0.0054544	Paxs	306.91	Joback Method
dvisc	0.0231540	Paxs	262.18	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R578653&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R578653&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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