

# 1-Penten-3-ol, acetate

<b>Inchi:</b>	InChI=1S/C7H12O2/c1-4-7(5-2)9-6(3)8/h4,7H,1,5H2,2-3H3
<b>InchiKey:</b>	MRLKTTBPWZXARX-UHFFFAOYSA-N
<b>Formula:</b>	C7H12O2
<b>SMILES:</b>	C=CC(CC)OC(C)=O
<b>Mol. weight [g/mol]:</b>	128.17

## Physical Properties

Property code	Value	Unit	Source
gf	-140.46	kJ/mol	Joback Method
hf	-312.46	kJ/mol	Joback Method
hfus	11.87	kJ/mol	Joback Method
hvap	39.27	kJ/mol	Joback Method
log10ws	-1.58		Crippen Method
logp	1.514		Crippen Method
mcvol	112.630	ml/mol	McGowan Method
pc	3107.10	kPa	Joback Method
rinpol	812.00		NIST Webbook
tb	432.09	K	Joback Method
tc	616.21	K	Joback Method
tf	224.05	K	Joback Method
vc	0.426	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	222.93	J/mol×K	432.09	Joback Method
cpg	233.73	J/mol×K	462.78	Joback Method
cpg	244.12	J/mol×K	493.46	Joback Method
cpg	254.10	J/mol×K	524.15	Joback Method
cpg	263.67	J/mol×K	554.84	Joback Method
cpg	272.85	J/mol×K	585.52	Joback Method
cpg	281.64	J/mol×K	616.21	Joback Method
dvisc	0.0042260	Paxs	224.05	Joback Method
dvisc	0.0019297	Paxs	258.72	Joback Method

dvisc	0.0010605	Paxs	293.40	Joback Method
dvisc	0.0006615	Paxs	328.07	Joback Method
dvisc	0.0004515	Paxs	362.74	Joback Method
dvisc	0.0003294	Paxs	397.42	Joback Method
dvisc	0.0002529	Paxs	432.09	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R98181&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R98181&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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