

# Phenol, 2,4'-isopropylidenedi-

**Other names:**

2,4'-Bisphenol A  
Phenol, 2-[1-(4-hydroxyphenyl)-1-methylethyl]-  
2,4'-Dihydroxydiphenyldimethylmethane  
2,4'-Isopropylidenediphenol  
2-(4'-Hydroxyphenyl)-2-(2'-hydroxyphenyl)propane  
o-[1-(4-hydroxyphenyl)-1-methylethyl]phenol

**Inchi:** InChI=1S/C15H16O2/c1-15(2,11-7-9-12(16)10-8-11)13-5-3-4-6-14(13)17/h3-10,16-17H,1**InchiKey:** MLCQXUZZAXKTSG-UHFFFAOYSA-N**Formula:** C15H16O2**SMILES:** CC(C)(c1ccc(O)cc1)c1ccccc1O**Mol. weight [g/mol]:** 228.29**CAS:** 837-08-1

## Physical Properties

Property code	Value	Unit	Source
gf	-6.16	kJ/mol	Joback Method
hf	-243.24	kJ/mol	Joback Method
hfus	26.84	kJ/mol	Joback Method
hvap	78.27	kJ/mol	Joback Method
log10ws	-3.18		Crippen Method
logp	3.424		Crippen Method
mcvol	186.430	ml/mol	McGowan Method
pc	3547.31	kPa	Joback Method
tb	753.97	K	Joback Method
tc	1018.62	K	Joback Method
tf	537.51	K	Joback Method
vc	0.581	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	528.97	J/mol×K	753.97	Joback Method
cpg	543.20	J/mol×K	798.08	Joback Method
cpg	556.63	J/mol×K	842.19	Joback Method

cpg	569.54	J/molxK	886.29	Joback Method
cpg	582.22	J/molxK	930.40	Joback Method
cpg	594.96	J/molxK	974.51	Joback Method
cpg	608.03	J/molxK	1018.62	Joback Method
dvisc	0.0000313	Paxs	537.51	Joback Method
dvisc	0.0000137	Paxs	573.59	Joback Method
dvisc	0.0000066	Paxs	609.66	Joback Method
dvisc	0.0000035	Paxs	645.74	Joback Method
dvisc	0.0000019	Paxs	681.82	Joback Method
dvisc	0.0000012	Paxs	717.89	Joback Method
dvisc	0.0000007	Paxs	753.97	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C837081&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C837081&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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