

# 1-Octene, 7-methyl-

<b>Other names:</b>	7-Methyl-1-octene
<b>Inchi:</b>	InChI=1S/C9H18/c1-4-5-6-7-8-9(2)3/h4,9H,1,5-8H2,2-3H3
<b>InchiKey:</b>	YKHFZRXXJMLNTJ-UHFFFAOYSA-N
<b>Formula:</b>	C9H18
<b>SMILES:</b>	C=CCCCC(C)C
<b>Mol. weight [g/mol]:</b>	126.24
<b>CAS:</b>	13151-06-9

## Physical Properties

Property code	Value	Unit	Source
gf	110.30	kJ/mol	Joback Method
hf	-108.94	kJ/mol	Joback Method
hfus	14.26	kJ/mol	Joback Method
hvap	34.57	kJ/mol	Joback Method
log10ws	-3.20		Crippen Method
logp	3.389		Crippen Method
mcvol	133.370	ml/mol	McGowan Method
pc	2421.88	kPa	Joback Method
rinpol	852.00		NIST Webbook
rinpol	847.00		NIST Webbook
rinpol	853.00		NIST Webbook
rinpol	847.00		NIST Webbook
rinpol	847.00		NIST Webbook
rinpol	850.00		NIST Webbook
rinpol	847.00		NIST Webbook
rinpol	847.00		NIST Webbook
tb	412.04 ± 0.40	K	NIST Webbook
tc	571.62	K	Joback Method
tf	174.43	K	Joback Method
vc	0.514	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	257.23	J/molxK	401.56	Joback Method
cpg	271.24	J/molxK	429.90	Joback Method
cpg	284.69	J/molxK	458.25	Joback Method
cpg	297.60	J/molxK	486.59	Joback Method
cpg	309.99	J/molxK	514.93	Joback Method
cpg	321.86	J/molxK	543.28	Joback Method
cpg	333.24	J/molxK	571.62	Joback Method
dvisc	0.0090007	Paxs	174.43	Joback Method
dvisc	0.0028336	Paxs	212.28	Joback Method
dvisc	0.0012657	Paxs	250.14	Joback Method
dvisc	0.0006987	Paxs	288.00	Joback Method
dvisc	0.0004428	Paxs	325.85	Joback Method
dvisc	0.0003086	Paxs	363.70	Joback Method
dvisc	0.0002302	Paxs	401.56	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C13151069&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C13151069&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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