

# Propanoic acid, 2,3-dichloro-

<b>Other names:</b>	Propionic acid, 2,3-dichloro- «alpha», «beta»-Dichloropropionic acid 2,3-Dichloropropionic acid Alpha,beta-dichloropropionic acid 2,3-Dichloropropanoic acid
<b>Inchi:</b>	InChI=1S/C3H4Cl2O2/c4-1-2(5)3(6)7/h2H,1H2,(H,6,7)
<b>InchiKey:</b>	GKFWNPPZHDYVLI-UHFFFAOYSA-N
<b>Formula:</b>	C3H4Cl2O2
<b>SMILES:</b>	O=C(O)C(Cl)CCl
<b>Mol. weight [g/mol]:</b>	142.97
<b>CAS:</b>	565-64-0

## Physical Properties

Property code	Value	Unit	Source
gf	-317.66	kJ/mol	Joback Method
hf	-406.82	kJ/mol	Joback Method
hfus	14.08	kJ/mol	Joback Method
hvap	54.08	kJ/mol	Joback Method
log10ws	-0.60		Crippen Method
logp	0.917		Crippen Method
mvol	85.050	ml/mol	McGowan Method
pc	4980.35	kPa	Joback Method
rinpol	1124.00		NIST Webbook
rinpol	1124.00		NIST Webbook
tb	488.51	K	Joback Method
tc	680.24	K	Joback Method
tf	323.00 ± 0.60	K	NIST Webbook
vc	0.321	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	148.88	J/mol×K	488.51	Joback Method
cpg	153.67	J/mol×K	520.46	Joback Method

cpg	158.21	J/mol×K	552.42	Joback Method
cpg	162.49	J/mol×K	584.37	Joback Method
cpg	166.52	J/mol×K	616.33	Joback Method
cpg	170.32	J/mol×K	648.28	Joback Method
cpg	173.89	J/mol×K	680.24	Joback Method
dvisc	0.0233353	Paxs	279.16	Joback Method
dvisc	0.0072444	Paxs	314.05	Joback Method
dvisc	0.0028417	Paxs	348.94	Joback Method
dvisc	0.0013215	Paxs	383.84	Joback Method
dvisc	0.0006981	Paxs	418.73	Joback Method
dvisc	0.0004069	Paxs	453.62	Joback Method
dvisc	0.0002561	Paxs	488.51	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C565640&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C565640&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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