

# Benzeneacetic acid, «alpha»-methyl-

<b>Other names:</b>	2-Phenylpropionic acid 2-Phenylpropanoic acid «alpha»-Phenylpropionic acid «alpha»-Methylphenylacetic acid Hydratropic acid dl-«alpha»-Phenylpropionic acid «alpha»-Methylbenzeneacetic acid Propanoic acid, 2-phenyl
<b>Inchi:</b>	InChI=1S/C9H10O2/c1-7(9(10)11)8-5-3-2-4-6-8/h2-7H,1H3,(H,10,11)
<b>InchiKey:</b>	YPGCWEMNNLXISK-UHFFFAOYSA-N
<b>Formula:</b>	C9H10O2
<b>SMILES:</b>	CC(C(=O)O)c1ccccc1
<b>Mol. weight [g/mol]:</b>	150.17
<b>CAS:</b>	492-37-5

## Physical Properties

Property code	Value	Unit	Source
gf	-130.87	kJ/mol	Joback Method
hf	-262.65	kJ/mol	Joback Method
hfus	15.27	kJ/mol	Joback Method
hvap	60.94	kJ/mol	Joback Method
log10ws	-1.76		Crippen Method
logp	1.875		Crippen Method
mcvol	121.350	ml/mol	McGowan Method
pc	3965.51	kPa	Joback Method
rinpol	1300.00		NIST Webbook
rinpol	1300.00		NIST Webbook
tb	534.20	K	NIST Webbook
tc	785.00	K	Joback Method
tf	313.36	K	Joback Method
vc	0.451	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	281.79	J/mol×K	577.61	Joback Method
cpg	329.12	J/mol×K	750.43	Joback Method
cpg	320.91	J/mol×K	715.87	Joback Method
cpg	312.10	J/mol×K	681.30	Joback Method
cpg	302.67	J/mol×K	646.74	Joback Method
cpg	292.57	J/mol×K	612.17	Joback Method
cpg	336.75	J/mol×K	785.00	Joback Method
dvisc	0.0001014	Paxs	577.61	Joback Method
dvisc	0.0001602	Paxs	533.57	Joback Method
dvisc	0.0002745	Paxs	489.53	Joback Method
dvisc	0.0005232	Paxs	445.49	Joback Method
dvisc	0.0011492	Paxs	401.44	Joback Method
dvisc	0.0030643	Paxs	357.40	Joback Method
dvisc	0.0107638	Paxs	313.36	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	424.50 ± 0.50	K	1.90	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C492375&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C492375&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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