

# 4-Methyl-2-hexanol

Inchi:	InChI=1S/C7H16O/c1-4-6(2)5-7(3)8/h6-8H,4-5H2,1-3H3
InchiKey:	KZUBXUKRWLMPIO-UHFFFAOYSA-N
Formula:	C7H16O
SMILES:	CCC(C)CC(C)O
Mol. weight [g/mol]:	116.20
CAS:	2313-61-3

## Physical Properties

Property code	Value	Unit	Source
gf	-133.64	kJ/mol	Joback Method
hf	-350.60	kJ/mol	Joback Method
hfus	10.93	kJ/mol	Joback Method
hvap	47.08	kJ/mol	Joback Method
log10ws	-1.89		Crippen Method
logp	1.803		Crippen Method
mvol	115.360	ml/mol	McGowan Method
pc	3149.09	kPa	Joback Method
tb	423.65 ± 1.50	K	NIST Webbook
tb	425.15 ± 2.00	K	NIST Webbook
tb	421.15 ± 3.00	K	NIST Webbook
tc	618.05	K	Joback Method
tf	199.47	K	Joback Method
vc	0.434	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	248.03	J/mol×K	450.86	Joback Method
cpg	299.70	J/mol×K	590.19	Joback Method
cpg	290.18	J/mol×K	562.32	Joback Method
cpg	280.27	J/mol×K	534.46	Joback Method
cpg	269.95	J/mol×K	506.59	Joback Method
cpg	259.20	J/mol×K	478.73	Joback Method
cpg	308.82	J/mol×K	618.05	Joback Method

dvisc	0.0001931	Paxs	450.86	Joback Method
dvisc	0.0003619	Paxs	408.96	Joback Method
dvisc	0.0007829	Paxs	367.06	Joback Method
dvisc	0.0020664	Paxs	325.16	Joback Method
dvisc	0.0072680	Paxs	283.27	Joback Method
dvisc	0.0395574	Paxs	241.37	Joback Method
dvisc	0.4386858	Paxs	199.47	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.49498e+01
Coeff. B	-3.06775e+03
Coeff. C	-1.24217e+02
Temperature range (K), min.	333.45
Temperature range (K), max.	442.50

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2313613&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2313613&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>p<sub>vap</sub>:</b>	Vapor pressure
<b>t<sub>b</sub>:</b>	Normal Boiling Point Temperature
<b>t<sub>c</sub>:</b>	Critical Temperature
<b>t<sub>f</sub>:</b>	Normal melting (fusion) point
<b>v<sub>c</sub>:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/13-500-1/4-Methyl-2-hexanol.pdf>

Generated by Cheméo on 2024-04-25 22:07:05.545625405 +0000 UTC m=+16372074.466202717.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.