

# Formic acid, 2-propylphenyl ester

<b>Inchi:</b>	InChI=1S/C10H12O2/c1-2-5-9-6-3-4-7-10(9)12-8-11/h3-4,6-8H,2,5H2,1H3
<b>InchiKey:</b>	KWSAMNKWMADWHG-UHFFFAOYSA-N
<b>Formula:</b>	C10H12O2
<b>SMILES:</b>	CCCc1ccccc1OC=O
<b>Mol. weight [g/mol]:</b>	164.20

## Physical Properties

Property code	Value	Unit	Source
gf	-68.42	kJ/mol	Joback Method
hf	-242.47	kJ/mol	Joback Method
hfus	18.79	kJ/mol	Joback Method
hvap	49.92	kJ/mol	Joback Method
log10ws	-2.59		Crippen Method
logp	2.174		Crippen Method
mvol	135.440	ml/mol	McGowan Method
pc	3069.34	kPa	Joback Method
rinpol	1249.00		NIST Webbook
rinpol	1249.00		NIST Webbook
tb	530.94	K	Joback Method
tc	739.58	K	Joback Method
tf	305.63	K	Joback Method
vc	0.522	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	301.37	J/molxK	530.94	Joback Method
cpg	314.28	J/molxK	565.71	Joback Method
cpg	326.52	J/molxK	600.49	Joback Method
cpg	338.10	J/molxK	635.26	Joback Method
cpg	349.03	J/molxK	670.03	Joback Method
cpg	359.33	J/molxK	704.80	Joback Method
cpg	369.00	J/molxK	739.58	Joback Method
dvisc	0.0019637	Paxs	305.63	Joback Method

dvisc	0.0011278	Paxs	343.18	Joback Method
dvisc	0.0007226	Paxs	380.73	Joback Method
dvisc	0.0005015	Paxs	418.29	Joback Method
dvisc	0.0003697	Paxs	455.84	Joback Method
dvisc	0.0002854	Paxs	493.39	Joback Method
dvisc	0.0002286	Paxs	530.94	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U368966&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U368966&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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