

# Isobutyric acid, 2,2,2-trichloroethyl ester

<b>Other names:</b>	2,2,2-Trichloroethyl isobutanoate Propanoic acid, 2-methyl, 2,2,2-trichloroethyl ester 2,2,2-Trichloroethyl 2-methylpropanoate
<b>Inchi:</b>	InChI=1S/C6H9Cl3O2/c1-4(2)5(10)11-3-6(7,8)9/h4H,3H2,1-2H3
<b>InchiKey:</b>	QNYXMOBLXOMWCE-UHFFFAOYSA-N
<b>Formula:</b>	C6H9Cl3O2
<b>SMILES:</b>	CC(C)C(=O)OCC(Cl)(Cl)Cl
<b>Mol. weight [g/mol]:</b>	219.49

## Physical Properties

Property code	Value	Unit	Source
gf	-269.67	kJ/mol	Joback Method
hf	-473.22	kJ/mol	Joback Method
hfus	15.74	kJ/mol	Joback Method
hvap	49.58	kJ/mol	Joback Method
log10ws	-2.52		Crippen Method
logp	2.556		Crippen Method
mcvol	139.560	ml/mol	McGowan Method
pc	2979.54	kPa	Joback Method
rinpol	1102.00		NIST Webbook
rinpol	1153.00		NIST Webbook
tb	521.59	K	Joback Method
tc	734.09	K	Joback Method
tf	306.72	K	Joback Method
vc	0.525	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	276.03	J/molxK	521.59	Joback Method
cpg	285.82	J/molxK	557.01	Joback Method
cpg	294.97	J/molxK	592.42	Joback Method
cpg	303.52	J/molxK	627.84	Joback Method
cpg	311.49	J/molxK	663.25	Joback Method

cpg	318.90	J/mol×K	698.67	Joback Method
cpg	325.79	J/mol×K	734.09	Joback Method
dvisc	0.0041126	Paxs	306.72	Joback Method
dvisc	0.0020546	Paxs	342.53	Joback Method
dvisc	0.0011705	Paxs	378.34	Joback Method
dvisc	0.0007350	Paxs	414.16	Joback Method
dvisc	0.0004970	Paxs	449.97	Joback Method
dvisc	0.0003560	Paxs	485.78	Joback Method
dvisc	0.0002670	Paxs	521.59	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U354634&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U354634&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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