

4-Hydroxy-6-oxohept-4-enoic acid, heptafluorobutyrate

Inchi:	InChI=1S/C11H9F7O5/c1-5(19)4-6(2-3-7(20)21)23-8(22)9(12,13)10(14,15)11(16,17)18/h
InchiKey:	WOLHNCOTKBYCDQ-XQRVVYSFSA-N
Formula:	C11H9F7O5
SMILES:	CC(=O)C=C(CCC(=O)O)OC(=O)C(F)(F)C(F)(F)C(F)(F)F
Mol. weight [g/mol]:	354.18

Physical Properties

Property code	Value	Unit	Source
gf	-1870.32	kJ/mol	Joback Method
hf	-2184.15	kJ/mol	Joback Method
hfus	32.53	kJ/mol	Joback Method
hvap	69.84	kJ/mol	Joback Method
log10ws	-3.31		Crippen Method
logp	2.700		Crippen Method
mcvol	190.390	ml/mol	McGowan Method
pc	2045.61	kPa	Joback Method
rinsol	1324.00		NIST Webbook
tb	716.53	K	Joback Method
tc	888.85	K	Joback Method
tf	438.92	K	Joback Method
vc	0.780	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	552.16	J/mol×K	716.53	Joback Method
cpg	560.54	J/mol×K	745.25	Joback Method
cpg	568.32	J/mol×K	773.97	Joback Method
cpg	575.54	J/mol×K	802.69	Joback Method
cpg	582.26	J/mol×K	831.41	Joback Method
cpg	588.52	J/mol×K	860.13	Joback Method
cpg	594.37	J/mol×K	888.85	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U374240&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvac:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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