

# 1,4-ethoxymethoxybutane

Inchi:	InChI=1S/C7H16O2/c1-3-9-7-5-4-6-8-2/h3-7H2,1-2H3
InchiKey:	DFCCXBQUXOSGFY-UHFFFAOYSA-N
Formula:	C7H16O2
SMILES:	CCOCCCCOC
Mol. weight [g/mol]:	132.20

## Physical Properties

Property code	Value	Unit	Source
gf	-201.94	kJ/mol	Joback Method
hf	-452.25	kJ/mol	Joback Method
hfus	16.26	kJ/mol	Joback Method
hvap	36.00	kJ/mol	Joback Method
log10ws	-0.93		Crippen Method
logp	1.450		Crippen Method
mcvol	121.230	ml/mol	McGowan Method
pc	2701.41	kPa	Joback Method
rinpol	873.70		NIST Webbook
rinpol	871.80		NIST Webbook
tb	404.40	K	Joback Method
tc	569.26	K	Joback Method
tf	213.11	K	Joback Method
vc	0.464	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	242.33	J/molxK	404.40	Joback Method
cpg	253.59	J/molxK	431.88	Joback Method
cpg	264.58	J/molxK	459.35	Joback Method
cpg	275.28	J/molxK	486.83	Joback Method
cpg	285.69	J/molxK	514.30	Joback Method
cpg	295.82	J/molxK	541.78	Joback Method
cpg	305.65	J/molxK	569.26	Joback Method
dvisc	0.0029460	Paxs	213.11	Joback Method

dvisc	0.0014040	Paxs	244.99	Joback Method
dvisc	0.0007936	Paxs	276.87	Joback Method
dvisc	0.0005047	Paxs	308.75	Joback Method
dvisc	0.0003493	Paxs	340.64	Joback Method
dvisc	0.0002575	Paxs	372.52	Joback Method
dvisc	0.0001992	Paxs	404.40	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R135914&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R135914&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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