

# Diethylmalonic acid, dodecyl 3-methylbenzyl ester

Inchi:	InChI=1S/C27H44O4/c1-5-8-9-10-11-12-13-14-15-16-20-30-25(28)27(6-2,7-3)26(29)31-2
InchiKey:	VRMHROITHIAZHK-UHFFFAOYSA-N
Formula:	C27H44O4
SMILES:	CCCCCCCCCCCCOC(=O)C(CC)(CC)C(=O)OCc1cccc(C)c1
Mol. weight [g/mol]:	432.64

## Physical Properties

Property code	Value	Unit	Source
gf	-185.76	kJ/mol	Joback Method
hf	-873.90	kJ/mol	Joback Method
hfus	57.50	kJ/mol	Joback Method
hvap	95.65	kJ/mol	Joback Method
log10ws	-8.26		Crippen Method
logp	7.309		Crippen Method
mcvol	382.410	ml/mol	McGowan Method
pc	873.77	kPa	Joback Method
rinsol	2849.00		NIST Webbook
tb	998.17	K	Joback Method
tc	1222.43	K	Joback Method
tf	579.73	K	Joback Method
vc	1.476	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1306.32	J/molxK	998.17	Joback Method
cpg	1381.47	J/molxK	1185.06	Joback Method
cpg	1369.08	J/molxK	1147.68	Joback Method
cpg	1355.45	J/molxK	1110.30	Joback Method
cpg	1340.50	J/molxK	1072.92	Joback Method
cpg	1324.15	J/molxK	1035.55	Joback Method
cpg	1392.70	J/molxK	1222.43	Joback Method
dvisc	0.0000148	Paxs	998.17	Joback Method
dvisc	0.0000197	Paxs	928.43	Joback Method

dvisc	0.0000274	Paxs	858.69	Joback Method
dvisc	0.0000406	Paxs	788.95	Joback Method
dvisc	0.0000649	Paxs	719.21	Joback Method
dvisc	0.0001146	Paxs	649.47	Joback Method
dvisc	0.0002323	Paxs	579.73	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369313&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369313&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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