

# Fumaric acid, pentachlorophenyl propyl ester

<b>Inchi:</b>	InChI=1S/C13H9Cl5O4/c1-2-5-21-6(19)3-4-7(20)22-13-11(17)9(15)8(14)10(16)12(13)18
<b>InchiKey:</b>	SZMBLVSWIKXQFX-ONEGZZNKSA-N
<b>Formula:</b>	C13H9Cl5O4
<b>SMILES:</b>	CCCOC(=O)C=CC(=O)Oc1c(Cl)c(Cl)c(Cl)c(Cl)c1Cl
<b>Mol. weight [g/mol]:</b>	406.47

## Physical Properties

Property code	Value	Unit	Source
gf	-324.43	kJ/mol	Joback Method
hf	-583.55	kJ/mol	Joback Method
hfus	48.28	kJ/mol	Joback Method
hvap	90.31	kJ/mol	Joback Method
log10ws	-6.02		Crippen Method
logp	5.368		Crippen Method
mcvol	242.050	ml/mol	McGowan Method
pc	1973.55	kPa	Joback Method
rinpol	2589.00		NIST Webbook
tb	892.31	K	Joback Method
tc	1126.39	K	Joback Method
tf	614.13	K	Joback Method
vc	0.928	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	562.69	J/molxK	892.31	Joback Method
cpg	570.47	J/molxK	931.32	Joback Method
cpg	577.39	J/molxK	970.34	Joback Method
cpg	583.47	J/molxK	1009.35	Joback Method
cpg	588.70	J/molxK	1048.36	Joback Method
cpg	593.09	J/molxK	1087.37	Joback Method
cpg	596.64	J/molxK	1126.39	Joback Method
dvisc	0.0002818	Paxs	614.13	Joback Method
dvisc	0.0002028	Paxs	660.49	Joback Method

dvisc	0.0001523	Paxs	706.86	Joback Method
dvisc	0.0001185	Paxs	753.22	Joback Method
dvisc	0.0000949	Paxs	799.58	Joback Method
dvisc	0.0000779	Paxs	845.95	Joback Method
dvisc	0.0000653	Paxs	892.31	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348179&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348179&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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