

# trans-methyl-i-butyl-diazene

Inchi:	InChI=1S/C5H12N2/c1-5(2)4-7-6-3/h5H,4H2,1-3H3/b7-6+
InchiKey:	MNPIONKINHESBM-VOTSOKGWSA-N
Formula:	C5H12N2
SMILES:	CN=NCC(C)C
Mol. weight [g/mol]:	100.16

## Physical Properties

Property code	Value	Unit	Source
hf	-104.59	kJ/mol	Joback Method
hvap	33.01	kJ/mol	Joback Method
log10ws	-0.93		Crippen Method
logp	1.724		Crippen Method
mcvol	96.970	ml/mol	McGowan Method
pc	2624.46	kPa	Joback Method
rinpol	642.20		NIST Webbook
tb	462.56	K	Joback Method
tc	668.86	K	Joback Method

## Sources

Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R166597&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R166597&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

hf:	Enthalpy of formation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l

<b>logP:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

Latest version available from:

<https://www.chemeo.com/cid/13-782-9/trans-methyl-i-butyl-diazene.pdf>

Generated by Cheméo on 2024-04-27 21:23:28.98959348 +0000 UTC m=+16542257.910170792.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.