

# Diethylmalonic acid, 4-biphenyl heptyl ester

<b>Inchi:</b>	InChI=1S/C26H34O4/c1-4-7-8-9-13-20-29-24(27)26(5-2,6-3)25(28)30-23-18-16-22(17-19
<b>InchiKey:</b>	DQDHEKWWONYPEK-UHFFFAOYSA-N
<b>Formula:</b>	C26H34O4
<b>SMILES:</b>	CCCCCCCOC(=O)C(CC)(CC)C(=O)Oc1ccc(-c2ccccc2)cc1
<b>Mol. weight [g/mol]:</b>	410.55

## Physical Properties

Property code	Value	Unit	Source
gf	-81.77	kJ/mol	Joback Method
hf	-616.73	kJ/mol	Joback Method
hfus	48.95	kJ/mol	Joback Method
hvap	95.70	kJ/mol	Joback Method
log10ws	-8.03		Crippen Method
logp	6.579		Crippen Method
mcvol	344.560	ml/mol	McGowan Method
pc	1144.44	kPa	Joback Method
rinsol	3021.00		NIST Webbook
tb	1001.97	K	Joback Method
tc	1231.42	K	Joback Method
tf	594.88	K	Joback Method
vc	1.312	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1137.91	J/molxK	1001.97	Joback Method
cpg	1199.62	J/molxK	1193.18	Joback Method
cpg	1189.62	J/molxK	1154.94	Joback Method
cpg	1178.53	J/molxK	1116.70	Joback Method
cpg	1166.28	J/molxK	1078.45	Joback Method
cpg	1152.77	J/molxK	1040.21	Joback Method
cpg	1208.65	J/molxK	1231.42	Joback Method
dvisc	0.0000178	Paxs	1001.97	Joback Method
dvisc	0.0000234	Paxs	934.12	Joback Method

dvisc	0.0000320	Paxs	866.27	Joback Method
dvisc	0.0000462	Paxs	798.42	Joback Method
dvisc	0.0000715	Paxs	730.58	Joback Method
dvisc	0.0001209	Paxs	662.73	Joback Method
dvisc	0.0002307	Paxs	594.88	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370427&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370427&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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