

# 3-Ethyl-5-methylbenzyl alcohol

<b>Inchi:</b>	InChI=1S/C10H14O/c1-3-9-4-8(2)5-10(6-9)7-11/h4-6,11H,3,7H2,1-2H3
<b>InchiKey:</b>	VITSERFATIPMIB-UHFFFAOYSA-N
<b>Formula:</b>	C10H14O
<b>SMILES:</b>	CCc1cc(C)cc(CO)c1
<b>Mol. weight [g/mol]:</b>	150.22

## Physical Properties

Property code	Value	Unit	Source
gf	-10.35	kJ/mol	Joback Method
hf	-188.37	kJ/mol	Joback Method
hfus	19.01	kJ/mol	Joback Method
hvap	58.13	kJ/mol	Joback Method
log10ws	-2.89		Crippen Method
logp	2.050		Crippen Method
mcvol	133.870	ml/mol	McGowan Method
pc	3138.51	kPa	Joback Method
rinpol	1219.00		NIST Webbook
rinpol	1219.00		NIST Webbook
tb	557.02	K	Joback Method
tc	751.47	K	Joback Method
tf	314.74	K	Joback Method
vc	0.506	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	311.00	J/molxK	557.02	Joback Method
cpg	364.99	J/molxK	719.07	Joback Method
cpg	355.31	J/molxK	686.66	Joback Method
cpg	345.08	J/molxK	654.25	Joback Method
cpg	334.31	J/molxK	621.84	Joback Method
cpg	322.95	J/molxK	589.43	Joback Method
cpg	374.16	J/molxK	751.47	Joback Method
dvisc	0.0000969	Paxs	557.02	Joback Method

dvisc	0.0001456	Paxs	516.64	Joback Method
dvisc	0.0002344	Paxs	476.26	Joback Method
dvisc	0.0004122	Paxs	435.88	Joback Method
dvisc	0.0008136	Paxs	395.50	Joback Method
dvisc	0.0018743	Paxs	355.12	Joback Method
dvisc	0.0053487	Paxs	314.74	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R553496&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R553496&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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