

# 1,3-Benzenedithiol

<b>Other names:</b>	Benzene-1,3-dithiol 1,3-Benzenedithiol
<b>Inchi:</b>	InChI=1S/C6H6S2/c7-5-2-1-3-6(8)4-5/h1-4,7-8H
<b>InchiKey:</b>	ZWOASCVFHSYHOB-UHFFFAOYSA-N
<b>Formula:</b>	C6H6S2
<b>SMILES:</b>	Sc1cccc(S)c1
<b>Mol. weight [g/mol]:</b>	142.24
<b>CAS:</b>	626-04-0

## Physical Properties

Property code	Value	Unit	Source
gf	161.20	kJ/mol	Joback Method
hf	134.85	kJ/mol	Joback Method
hfus	13.03	kJ/mol	Joback Method
hvap	45.36	kJ/mol	Joback Method
log10ws	-2.50		Crippen Method
logp	2.264		Crippen Method
mvol	104.340	ml/mol	McGowan Method
pc	5486.97	kPa	Joback Method
ripol	2467.00		NIST Webbook
ripol	2467.00		NIST Webbook
tb	518.00	K	NIST Webbook
tb	518.20	K	NIST Webbook
tc	766.38	K	Joback Method
tf	300.00	K	NIST Webbook
vc	0.371	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	182.33	J/mol×K	494.06	Joback Method
cpg	192.58	J/mol×K	539.45	Joback Method
cpg	202.02	J/mol×K	584.83	Joback Method
cpg	210.70	J/mol×K	630.22	Joback Method

cpg	218.65	J/mol×K	675.61	Joback Method
cpg	225.92	J/mol×K	721.00	Joback Method
cpg	232.55	J/mol×K	766.38	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	401.70	K	2.10	NIST Webbook
tbrp	335.00	K	0.03	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C626040&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C626040&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/13-905-2/1-3-Benzenedithiol.pdf>

Generated by Cheméo on 2024-05-02 10:43:16.430779553 +0000 UTC m=+16935845.351356865.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.