

# Acetic acid, tribromo, 2-methylpropyl ester

<b>Inchi:</b>	InChI=1S/C6H9Br3O2/c1-4(2)3-11-5(10)6(7,8)9/h4H,3H2,1-2H3
<b>InchiKey:</b>	GBRCAOJHYDQQGKG-UHFFFAOYSA-N
<b>Formula:</b>	C6H9Br3O2
<b>SMILES:</b>	CC(C)COC(=O)C(Br)(Br)Br
<b>Mol. weight [g/mol]:</b>	352.85

## Physical Properties

Property code	Value	Unit	Source
gf	-190.92	kJ/mol	Joback Method
hf	-347.01	kJ/mol	Joback Method
hfus	19.00	kJ/mol	Joback Method
hvap	55.73	kJ/mol	Joback Method
log10ws	-3.35		Crippen Method
logp	3.024		Crippen Method
mcvol	155.340	ml/mol	McGowan Method
pc	4362.63	kPa	Joback Method
rinpol	1387.00		NIST Webbook
tb	607.78	K	Joback Method
tc	848.93	K	Joback Method
tf	396.36	K	Joback Method
vc	0.565	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	302.29	J/molxK	607.78	Joback Method
cpg	340.54	J/molxK	808.74	Joback Method
cpg	334.12	J/molxK	768.55	Joback Method
cpg	327.15	J/molxK	728.36	Joback Method
cpg	319.57	J/molxK	688.16	Joback Method
cpg	311.30	J/molxK	647.97	Joback Method
cpg	346.48	J/molxK	848.93	Joback Method
dvisc	0.0002206	Paxs	607.78	Joback Method
dvisc	0.0002807	Paxs	572.54	Joback Method

dvisc	0.0003685	Paxs	537.31	Joback Method
dvisc	0.0005026	Paxs	502.07	Joback Method
dvisc	0.0007184	Paxs	466.83	Joback Method
dvisc	0.0010885	Paxs	431.60	Joback Method
dvisc	0.0017759	Paxs	396.36	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R115821&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R115821&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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