

# 3-Methyl-4-hexyn-3-ol

|                      |                                                |
|----------------------|------------------------------------------------|
| Inchi:               | InChI=1S/C7H12O/c1-4-6-7(3,8)5-2/h8H,5H2,1-3H3 |
| InchiKey:            | VLRDHNINMBNPDK-UHFFFAOYSA-N                    |
| Formula:             | C7H12O                                         |
| SMILES:              | CC#CC(C)(O)CC                                  |
| Mol. weight [g/mol]: | 112.17                                         |
| CAS:                 | 6320-68-9                                      |

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | 76.88   | kJ/mol  | Joback Method  |
| hf            | -76.49  | kJ/mol  | Joback Method  |
| hfus          | 13.68   | kJ/mol  | Joback Method  |
| hvap          | 48.71   | kJ/mol  | Joback Method  |
| log10ws       | -1.92   |         | Crippen Method |
| logp          | 1.171   |         | Crippen Method |
| mcvol         | 106.760 | ml/mol  | McGowan Method |
| pc            | 3815.10 | kPa     | Joback Method  |
| tb            | 457.51  | K       | Joback Method  |
| tc            | 650.27  | K       | Joback Method  |
| tf            | 337.99  | K       | Joback Method  |
| vc            | 0.398   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 220.44 | J/mol×K | 457.51          | Joback Method |
| cpg           | 230.91 | J/mol×K | 489.64          | Joback Method |
| cpg           | 240.81 | J/mol×K | 521.76          | Joback Method |
| cpg           | 250.19 | J/mol×K | 553.89          | Joback Method |
| cpg           | 259.05 | J/mol×K | 586.01          | Joback Method |
| cpg           | 267.43 | J/mol×K | 618.14          | Joback Method |
| cpg           | 275.36 | J/mol×K | 650.27          | Joback Method |

# Correlations

| Information                 | Value                         |
|-----------------------------|-------------------------------|
| Property code               | pvap                          |
| Equation                    | $\ln(P_{vp}) = A + B/(T + C)$ |
| Coeff. A                    | 1.64196e+01                   |
| Coeff. B                    | -4.55664e+03                  |
| Coeff. C                    | -6.71610e+01                  |
| Temperature range (K), min. | 349.15                        |
| Temperature range (K), max. | 477.15                        |

## Sources

|                                      |                                                                                                                                                                                         |
|--------------------------------------|-----------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| Crippen Method:                      | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                                                                       |
| Joback Method:                       | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                                                                   |
| McGowan Method:                      | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                                                                   |
| NIST Webbook:                        | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6320689&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6320689&amp;Units=SI</a>                                             |
| The Yaws Handbook of Vapor Pressure: | <a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a> |
| Crippen Method:                      | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                                                               |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>pvap:</b>    | Vapor pressure                                  |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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