

# Benzoic acid, 4-hydroxy-, hydrazide

<b>Other names:</b>	Benzoic acid, p-hydroxy-, hydrazide (p-Hydroxybenzoyl)hydrazine (4-Hydroxybenzoyl)hydrazine p-Hydroxybenzhydrazide p-Hydroxybenzoic acid hydrazide p-Hydroxybenzoic hydrazide 4-Hydroxybenzohydrazide 4-Hydroxybenzoic acid hydrazide para-Hydroxybenzoic acid hydrazide 4-Hydroxybenzhydrazide NSC 647 p-Hydroxybenzoyl hydrazide
<b>Inchi:</b>	InChI=1S/C7H8N2O2/c8-9-7(11)5-1-3-6(10)4-2-5/h1-4,10H,8H2,(H,9,11)
<b>InchiKey:</b>	ZMZGIVVRBMFZSG-UHFFFAOYSA-N
<b>Formula:</b>	C7H8N2O2
<b>SMILES:</b>	NNC(=O)c1ccc(O)cc1
<b>Mol. weight [g/mol]:</b>	152.15
<b>CAS:</b>	5351-23-5

## Physical Properties

Property code	Value	Unit	Source
gf	-7.23	kJ/mol	Joback Method
hf	-153.91	kJ/mol	Joback Method
hfus	25.61	kJ/mol	Joback Method
hvap	70.29	kJ/mol	Joback Method
log10ws	-1.37		Crippen Method
logp	-0.004		Crippen Method
mcvol	113.130	ml/mol	McGowan Method
pc	6132.23	kPa	Joback Method
tb	643.43	K	Joback Method
tc	889.21	K	Joback Method
tf	492.64	K	Joback Method
vc	0.355	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	283.79	J/mol×K	643.43	Joback Method
cpg	292.93	J/mol×K	684.39	Joback Method
cpg	301.33	J/mol×K	725.36	Joback Method
cpg	309.10	J/mol×K	766.32	Joback Method
cpg	316.34	J/mol×K	807.28	Joback Method
cpg	323.14	J/mol×K	848.25	Joback Method
cpg	329.62	J/mol×K	889.21	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5351235&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5351235&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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