

3-Decanone

Other names:	Decan-3-one Ethyl heptyl ketone Ethyl n-heptyl ketone
Inchi:	InChI=1S/C10H20O/c1-3-5-6-7-8-9-10(11)4-2/h3-9H2,1-2H3
InchiKey:	XJLDYKIEURAVBW-UHFFFAOYSA-N
Formula:	C10H20O
SMILES:	CCCCCCCCC(=O)CC
Mol. weight [g/mol]:	156.27
CAS:	928-80-3

Physical Properties

Property code	Value	Unit	Source
gf	-95.60	kJ/mol	Joback Method
hf	-362.31	kJ/mol	Joback Method
hfus	23.25	kJ/mol	Joback Method
hvap	44.60	kJ/mol	Joback Method
log10ws	-3.29		Crippen Method
logp	3.326		Crippen Method
mcvol	153.330	ml/mol	McGowan Method
pc	2235.52	kPa	Joback Method
rhoc	248.46 ± 6.25	kg/m3	NIST Webbook
rinpol	1168.00		NIST Webbook
rinpol	1184.00		NIST Webbook
rinpol	1184.00		NIST Webbook
rinpol	1188.00		NIST Webbook
rinpol	1165.00		NIST Webbook
rinpol	1165.00		NIST Webbook
rinpol	1184.00		NIST Webbook
rinpol	1176.00		NIST Webbook
rinpol	1187.00		NIST Webbook
rinpol	1177.00		NIST Webbook
rinpol	1165.00		NIST Webbook
rinpol	1168.00		NIST Webbook
rinpol	1195.00		NIST Webbook
ripol	1455.00		NIST Webbook
ripol	1455.00		NIST Webbook
ripol	1491.00		NIST Webbook

ripol	1455.00		NIST Webbook
tb	484.00 ± 6.00	K	NIST Webbook
tb	477.70	K	NIST Webbook
tb	478.00 ± 3.00	K	NIST Webbook
tc	667.60 ± 0.50	K	NIST Webbook
tf	275.60 ± 4.00	K	NIST Webbook
vc	0.602	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	342.00	J/mol×K	482.07	Joback Method
cpg	356.42	J/mol×K	510.87	Joback Method
cpg	370.27	J/mol×K	539.66	Joback Method
cpg	383.54	J/mol×K	568.46	Joback Method
cpg	396.27	J/mol×K	597.26	Joback Method
cpg	408.46	J/mol×K	626.06	Joback Method
cpg	420.12	J/mol×K	654.85	Joback Method
dvisc	0.0045381	Pa×s	252.39	Joback Method
dvisc	0.0020766	Pa×s	290.67	Joback Method
dvisc	0.0011399	Pa×s	328.95	Joback Method
dvisc	0.0007090	Pa×s	367.23	Joback Method
dvisc	0.0004824	Pa×s	405.51	Joback Method
dvisc	0.0003508	Pa×s	443.79	Joback Method
dvisc	0.0002683	Pa×s	482.07	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.49623e+01
Coeff. B	-4.18082e+03
Coeff. C	-7.35200e+01
Temperature range (K), min.	358.42
Temperature range (K), max.	506.73

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C928803&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure

Legend

cp_g:	Ideal gas heat capacity
dv_{isc}:	Dynamic viscosity
g_f:	Standard Gibbs free energy of formation
h_f:	Enthalpy of formation at standard conditions
h_{fus}:	Enthalpy of fusion at standard conditions
h_{vap}:	Enthalpy of vaporization at standard conditions
log_{10ws}:	Log10 of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
mc_{vol}:	McGowan's characteristic volume
p_c:	Critical Pressure
p_{vap}:	Vapor pressure
ρ_{hc}:	Critical density
ri_{npol}:	Non-polar retention indices
ri_{pol}:	Polar retention indices
t_b:	Normal Boiling Point Temperature
t_c:	Critical Temperature
t_f:	Normal melting (fusion) point
v_c:	Critical Volume

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