## 1,7-Heptanediol

Other names: .alpha.,.omega.-heptanediol

.omega.-heptanediol 1,7-Dihydroxyheptane Heptamethylene glycol

Heptane-1,7-diol

«alpha», «omega»-Heptanediol

«omega»-Heptanediol

«alpha»,«omega»-Heptanediol

«omega»-Heptanediol

InChl=1S/C7H16O2/c8-6-4-2-1-3-5-7-9/h8-9H,1-7H2

InchiKey: SXCBDZAEHILGLM-UHFFFAOYSA-N

Formula: C7H16O2 SMILES: OCCCCCCO

Mol. weight [g/mol]: 132.20 CAS: 629-30-1

### **Physical Properties**

Property code	Value	Unit	Source
chl	-4467.00 ± 9.30	kJ/mol	NIST Webbook
gf	-265.58	kJ/mol	Joback Method
hf	-477.60 ± 9.30	kJ/mol	NIST Webbook
hfl	-574.20 ± 9.30	kJ/mol	NIST Webbook
hfus	22.06	kJ/mol	Joback Method
hvap	$96.60 \pm 0.60$	kJ/mol	NIST Webbook
hvap	$96.50 \pm 3.20$	kJ/mol	NIST Webbook
hvap	96.20 ± 1.20	kJ/mol	NIST Webbook
hvap	96.60	kJ/mol	NIST Webbook
log10ws	-1.28		Crippen Method
logp	0.922		Crippen Method
mcvol	121.230	ml/mol	McGowan Method
рс	3435.91	kPa	Joback Method
rinpol	1201.00		NIST Webbook
rinpol	1201.00		NIST Webbook
tb	535.20	K	NIST Webbook
tb	532.20	K	NIST Webbook
tc	700.93	K	Joback Method
tf	293.15 ± 4.00	K	NIST Webbook

tf	$337.15 \pm 2.00$	K	NIST Webbook
tf	290.50	К	Thermodynamics of fusion and sublimation for a homologous series of eleven alkanealpha.,.omegadiols HO-(CH2)n-OH: Structure-related odd even effect
tt	$295.00 \pm 0.20$	K	NIST Webbook
VC	0.466	m3/kmol	Joback Method

# **Temperature Dependent Properties**

Property code	Value	Unit	Temperature [I	K] Source
cpg	317.33	J/mol×K	596.26	Joback Method
cpg	349.86	J/mol×K	700.93	Joback Method
cpg	342.22	J/mol×K	674.76	Joback Method
cpg	334.25	J/mol×K	648.59	Joback Method
cpg	325.96	J/mol×K	622.42	Joback Method
cpg	308.35	J/mol×K	570.09	Joback Method
cpg	299.02	J/mol×K	543.92	Joback Method
cpl	354.42	J/mol×K	353.15 a	Heat Capacities of Some Liquid Ipha,omega-Alkanedid within the Temperature Range between (293.15 and 353.15) K
cpl	342.20	J/mol×K	342.65 a	Heat Capacities of Some Liquid Ipha,omega-Alkanedid within the Temperature Range between (293.15 and 353.15) K
cpl	340.50	J/mol×K	341.15 a	Heat Capacities of Some Liquid Ipha,omega-Alkanedid within the Temperature Range between (293.15 and 353.15) K

cpl	338.81	J/mol×K	339.65	Heat Capacities of Some Liquid alpha,omega-Alkanediols
				within the Temperature Range between (293.15 and 353.15) K
cpl	337.13	J/mol×K	338.15	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	343.91	J/mol×K	344.15	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	345.63	J/mol×K	345.65	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	335.47	J/mol×K	336.65	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	347.37	J/mol×K	347.15	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	349.11	J/mol×K	348.65	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K

cpl	350.87	J/mol×K	350.15	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	352.64	J/mol×K	351.65	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	292.06	J/mol×K	293.15	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	293.40	J/mol×K	294.65	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	294.75	J/mol×K	296.15	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	296.11	J/mol×K	297.65	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	297.48	J/mol×K	299.15	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K

cpl	298.87	J/mol×K	300.65	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	300.26	J/mol×K	302.15	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	301.67	J/mol×K	303.65	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	303.09	J/mol×K	305.15	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	304.51	J/mol×K	306.65	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	305.95	J/mol×K	308.15	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	332.17	J/mol×K	333.65	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K

cpl	308.87	J/mol×K	311.15	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	310.34	J/mol×K	312.65	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	311.83	J/mol×K	314.15	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	313.33	J/mol×K	315.65	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	314.83	J/mol×K	317.15	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	316.35	J/mol×K	318.65	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	317.88	J/mol×K	320.15	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K

cpl	319.43	J/mol×K	321.65	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	320.98	J/mol×K	323.15	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	322.54	J/mol×K	324.65	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	324.12	J/mol×K	326.15	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	325.71	J/mol×K	327.65	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	327.31	J/mol×K	329.15	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	328.00	J/mol×K	330.65	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K

cpl	330.54	J/mol×K	332.15	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	307.41	J/mol×K	309.65	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
cpl	333.81	J/mol×K	335.15	Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Temperature Range between (293.15 and 353.15) K
dvisc	0.0001190	Paxs	501.65	Joback Method
dvisc	0.0002672	Paxs	459.38	Joback Method
dvisc	0.0007068	Paxs	417.11	Joback Method
dvisc	0.0023279	Paxs	374.83	Joback Method
dvisc	0.0103813	Paxs	332.56	Joback Method
dvisc	0.0000601	Paxs	543.92	Joback Method
dvisc	0.0715544	Paxs	290.29	Joback Method
hfust	21.30	kJ/mol	295.20	NIST Webbook
hfust	21.30	kJ/mol	295.20	NIST Webbook
hvapt	93.80	kJ/mol	323.00	NIST Webbook
hvapt	92.40	kJ/mol	341.00	NIST Webbook
hvapt	97.90	kJ/mol	298.15	Vaporization Enthalpies of the r,o-Alkanediols by Correlation Gas Chromatography

## **Correlations**

Information Value

Property code	pvap
Equation	In(Pvp) = A + B/(T + C)
Coeff. A	1.66473e+01
Coeff. B	-5.32013e+03
Coeff. C	-8.99220e+01

559.24

#### **Sources**

McGowan Method:http://link.springer.com/article/10.1007/BF02311772Crippen Method:https://www.chemeo.com/doc/models/crippen\_log10ws

Thermodynamics of fusion and sublimation for a homologous series of MSVeWarkanekalpha.,.omega.-diols HO-(CH2)n-OH: Structure-related odd then ewec Handbook of Vapor Pressure:
Heat Capacities of Some Liquid

Heat Capacities of Some Liquid alpha,omega-Alkanediols within the Feinpera Math Relinge between (293.15

and 353 15) K: Joback Method:

Vaporization Enthalpies of the r,o-Alkanediols by Correlation Gas Chromatography:

https://www.doi.org/10.1016/j.jct.2013.08.019

http://webbook.nist.gov/cgi/cbook.cgi?ID=C629301&Units=SI

https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure

https://www.doi.org/10.1021/je800356x

http://pubs.acs.org/doi/abs/10.1021/ci990307l https://en.wikipedia.org/wiki/Joback\_method

https://www.doi.org/10.1021/je060333x

#### Legend

**chl:** Standard liquid enthalpy of combustion

cpg: Ideal gas heat capacitycpl: Liquid phase heat capacity

**dvisc:** Dynamic viscosity

gf: Standard Gibbs free energy of formationhf: Enthalpy of formation at standard conditions

hfl: Liquid phase enthalpy of formation at standard conditions

hfus: Enthalpy of fusion at standard conditions hfust: Enthalpy of fusion at a given temperature

hvap: Enthalpy of vaporization at standard conditions hvapt: Enthalpy of vaporization at a given temperature

log10ws: Log10 of Water solubility in mol/llogp: Octanol/Water partition coefficientmcvol: McGowan's characteristic volume

pc: Critical Pressurepvap: Vapor pressure

rinpol: Non-polar retention indices

**tb:** Normal Boiling Point Temperature

tc: Critical Temperature

tf: Normal melting (fusion) pointtt: Triple Point Temperature

vc: Critical Volume

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