

# 1,3-Dithiane, 2-methyl-

<b>Other names:</b>	m-Dithiane, 2-methyl- 2-Methyl-1,3-dithiane 2-Methyl-m-dithiane Methyl-2 dithiane-1,3 Acetaldehyde, cyclic trimethylene mercaptal
<b>Inchi:</b>	InChI=1S/C5H10S2/c1-5-6-3-2-4-7-5/h5H,2-4H2,1H3
<b>InchiKey:</b>	KXROTPXCYDXGSC-UHFFFAOYSA-N
<b>Formula:</b>	C5H10S2
<b>SMILES:</b>	CC1SCCCS1
<b>Mol. weight [g/mol]:</b>	134.26
<b>CAS:</b>	6007-26-7

## Physical Properties

Property code	Value	Unit	Source
gf	95.39	kJ/mol	Joback Method
hf	-1.69	kJ/mol	Joback Method
hfus	7.85	kJ/mol	Joback Method
hvap	38.78	kJ/mol	Joback Method
log10ws	-2.18		Crippen Method
logp	2.202		Crippen Method
mcvol	103.150	ml/mol	McGowan Method
pc	4414.96	kPa	Joback Method
rinpol	1033.00		NIST Webbook
rinpol	1023.00		NIST Webbook
rinpol	1029.00		NIST Webbook
rinpol	1024.00		NIST Webbook
rinpol	1019.00		NIST Webbook
rinpol	1035.00		NIST Webbook
rinpol	1033.00		NIST Webbook
rinpol	1023.00		NIST Webbook
ripol	1480.00		NIST Webbook
ripol	1476.00		NIST Webbook
tb	429.01	K	Joback Method
tc	672.02	K	Joback Method
tf	320.39	K	Joback Method
vc	0.341	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	185.03	J/mol×K	429.01	Joback Method
cpg	199.05	J/mol×K	469.51	Joback Method
cpg	212.21	J/mol×K	510.01	Joback Method
cpg	224.55	J/mol×K	550.52	Joback Method
cpg	236.10	J/mol×K	591.02	Joback Method
cpg	246.88	J/mol×K	631.52	Joback Method
cpg	256.93	J/mol×K	672.02	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	330.70	K	0.40	NIST Webbook

## Sources

Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6007267&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6007267&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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