

# 2-[2-(2-Ethoxyethoxy)ethoxy]ethyl 2,2,3,3,4,4,4-heptafluorobutanoate

Other names:

Triethylene glycol monoethyl ether, heptafluorobutyrate

3,6,9-Trioxaundec-1-yl heptafluorobutyrate

Inchi:

InChI=1S/C12H17F7O5/c1-2-21-3-4-22-5-6-23-7-8-24-9(20)10(13,14)11(15,16)12(17,18)

InchiKey:

IMIBMDPKWPTZHS-UHFFFAOYSA-N

Formula:

C12H17F7O5

SMILES:

CCOCCOCCOCCOC(=O)C(F)(F)C(F)(F)C(F)(F)F

Mol. weight [g/mol]:

374.25

## Physical Properties

Property code	Value	Unit	Source
gf	-1853.91	kJ/mol	Joback Method
hf	-2331.49	kJ/mol	Joback Method
hfus	32.50	kJ/mol	Joback Method
hvap	49.09	kJ/mol	Joback Method
log10ws	-2.26		Crippen Method
logp	2.432		Crippen Method
mcvol	217.380	ml/mol	McGowan Method
pc	1442.44	kPa	Joback Method
rinpol	1327.50		NIST Webbook
tb	602.71	K	Joback Method
tc	754.24	K	Joback Method
tf	375.24	K	Joback Method
vc	0.878	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	613.27	J/molxK	602.71	Joback Method
cpg	626.47	J/molxK	627.97	Joback Method
cpg	639.03	J/molxK	653.22	Joback Method
cpg	650.97	J/molxK	678.48	Joback Method
cpg	662.31	J/molxK	703.73	Joback Method
cpg	673.05	J/molxK	728.99	Joback Method
cpg	683.23	J/molxK	754.24	Joback Method

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U352017&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U352017&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvpap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinppl:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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