# Nonane, 1-bromo-

Other names: 1-Bromononane

1-NONYL BROMIDE 1-n-Nonyl bromide N-NONYL BROMIDE

Nonyl bromide

n-Nonyl-1-bromide

**InChl**=1S/C9H19Br/c1-2-3-4-5-6-7-8-9-10/h2-9H2,1H3

InchiKey: AYMUQTNXKPEMLM-UHFFFAOYSA-N

Formula: C9H19Br

SMILES: CCCCCCCCBr

Mol. weight [g/mol]: 207.15 CAS: 693-58-3

#### **Physical Properties**

Property code	Value	Unit	Source
gf	39.22	kJ/mol	Joback Method
hf	-202.76	kJ/mol	Joback Method
hfus	24.35	kJ/mol	Joback Method
hvap	42.06	kJ/mol	Joback Method
log10ws	-4.02		Crippen Method
logp	4.132		Crippen Method
mcvol	155.170	ml/mol	McGowan Method
рс	2467.81	kPa	Joback Method
rinpol	1238.00		NIST Webbook
rinpol	1231.00		NIST Webbook
rinpol	1244.00		NIST Webbook
rinpol	1240.00		NIST Webbook
rinpol	1231.00		NIST Webbook
rinpol	1220.00		NIST Webbook
rinpol	1220.00		NIST Webbook
ripol	1478.00		NIST Webbook
ripol	1476.00		NIST Webbook
ripol	1496.00		NIST Webbook
ripol	1496.00		NIST Webbook
tb	474.20	K	NIST Webbook
tb	474.00	K	NIST Webbook
tc	650.81	K	Joback Method

tf	244.15	K	KDB
tf	$244.20 \pm 0.20$	K	NIST Webbook
tf	242.40 ± 0.20	K	NIST Webbook
tf	$244.09 \pm 0.25$	K	NIST Webbook
tf	242.44 ± 0.15	K	NIST Webbook
VC	0.602	m3/kmol	Joback Method

## **Temperature Dependent Properties**

cpl	324.63	J/mol×K	334.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis
cpl	310.56	J/mol×K	303.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes
cpl	323.32	J/mol×K	331.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis
cpl	326.99	J/mol×K	339.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis
cpl	328.13	J/mol×K	341.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis

cpl	329.39	J/mol×K	344.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	330.49	J/mol×K	346.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	331.79	J/mol×K	349.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	333.04	J/mol×K	351.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	

cpl	333.74	J/mol×K	353.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	308.30	J/mol×K	298.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	
cpl	322.16	J/mol×K	329.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	312.82	J/mol×K	308.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes	
cpl	315.10	J/mol×K	313.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	
cpl	317.36	J/mol×K	318.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	
cpl	319.61	J/mol×K	323.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	

cpl	321.87	J/mol×K	328.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	
cpl	324.15	J/mol×K	333.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes	
cpl	326.41	J/mol×K	338.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	
cpl	328.67	J/mol×K	343.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes	
cpl	330.92	J/mol×K	348.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	
cpl	333.20	J/mol×K	353.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	
cpl	335.46	J/mol×K	358.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	
cpl	337.72	J/mol×K	363.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	
cpl	339.98	J/mol×K	368.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	

срІ	342.26	J/mol×K	373.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes	
cpl	344.51	J/mol×K	378.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	
cpl	346.77	J/mol×K	383.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes	
cpl	349.03	J/mol×K	388.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes	
cpl	351.31	J/mol×K	393.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes	
cpl	353.57	J/mol×K	398.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes	
cpl	320.96	J/mol×K	326.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	358.08	J/mol×K	408.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1-bromoalkanes	

cpl	360.36	J/mol×K	413.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes	
cpl	362.62	J/mol×K	418.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes	
cpl	364.88	J/mol×K	423.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes	
cpl	302.55	J/mol×K	284.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	303.54	J/mol×K	286.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	304.62	J/mol×K	289.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	

cpl	305.57	J/mol×K	291.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	306.69	J/mol×K	294.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	307.68	J/mol×K	296.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	308.80	J/mol×K	299.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	

cpl	309.82	J/mol×K	301.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	310.95	J/mol×K	304.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	311.91	J/mol×K	306.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	313.01	J/mol×K	309.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	

cpl	314.10	J/mol×K	311.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	315.18	J/mol×K	314.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	316.32	J/mol×K	316.65	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	317.44	J/mol×K	319.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
cpl	355.82	J/mol×K	403.15	Isobaric heat capacity, isothermal compressibility and fluctuational properties of 1- bromoalkanes	

cpl	319.80	J/mol×K	324.15	Heat Capacities of 1-chloroalkanes and 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis	
dvisc	0.0002865	Paxs	471.48	Joback Method	
dvisc	0.0003711	Paxs	434.73	Joback Method	
dvisc	0.0005042	Paxs	397.98	Joback Method	
dvisc	0.0007290	Paxs	361.24	Joback Method	
dvisc	0.0011459	Paxs	324.49	Joback Method	
dvisc	0.0020219	Paxs	287.74	Joback Method	
dvisc	0.0042127	Paxs	250.99	Joback Method	
hfust	30.12	kJ/mol	243.20	NIST Webbook	
hfust	30.12	kJ/mol	243.20	NIST Webbook	
hvapt	53.10	kJ/mol	450.50	NIST Webbook	
hvapt	52.20	kJ/mol	470.00	NIST Webbook	
speedsl	1329.91	m/s	253.46	Speed of Sound, Densities, and Isentropic Compressibilities of Liquid 1-Bromoalkanes at Temperatures from (243.15 to 423.15) K	
speedsl	1262.72	m/s	273.29	Speed of Sound, Densities, and Isentropic Compressibilities of Liquid 1-Bromoalkanes at Temperatures from (243.15 to 423.15) K	_
speedsl	1101.12	m/s	323.14	Speed of Sound, Densities, and Isentropic Compressibilities of Liquid 1-Bromoalkanes at Temperatures from (243.15 to 423.15) K	

speedsl	1009.85	m/s	353.10	Speed of Sound, Densities, and Isentropic Compressibilities of Liquid 1-Bromoalkanes at Temperatures from (243.15 to 423.15) K
speedsl	921.46	m/s	383.05	Speed of Sound, Densities, and Isentropic Compressibilities of Liquid 1-Bromoalkanes at Temperatures from (243.15 to 423.15) K
speedsl	808.79	m/s	422.72	Speed of Sound, Densities, and Isentropic Compressibilities of Liquid 1-Bromoalkanes at Temperatures from (243.15 to 423.15) K
speedsl	864.42	m/s	402.98	Speed of Sound, Densities, and Isentropic Compressibilities of Liquid 1-Bromoalkanes at Temperatures from (243.15 to 423.15) K
speedsl	1165.07	m/s	303.15	Speed of Sound, Densities, and Isentropic Compressibilities of Liquid 1-Bromoalkanes at Temperatures from (243.15 to 423.15) K

### **Correlations**

Information Value

Property code	pvap
Equation	In(Pvp) = A + B/(T + C)
Coeff. A	1.76688e+01
Coeff. B	-5.16229e+03
Coeff. C	-7.86350e+01
Temperature range (K), min.	375.64

Information

Value

Property code	pvap
Equation	$ln(Pvp) = A + B/T + C*ln(T) + D*T^2$
Coeff. A	9.69072e+01
Coeff. B	-1.03437e+04
Coeff. C	-1.17047e+01
Coeff. D	5.06259e-06
Temperature range (K), min.	391.15
Temperature range (K), max.	549.15

#### Sources

**Crippen Method:** http://pubs.acs.org/doi/abs/10.1021/ci990307l

Speed of Sound, Densities, and entropic Compressibilities of Liquid KBRomoalkanes at Temperatures from

(243.15 to 423.15) K: McGowan Method:

Joback Method:

The Yaws Handbook of Vapor

Pressure: NIST Webbook:

**Crippen Method:** 

**KDB Vapor Pressure Data:** 

Isobaric heat capacity, isothermal compressibility and fluctuational ปู่คระผลองที่เจราจากปกเลลประการ and https://www.doi.org/10.1021/je049652j 1-bromoalkanes within the temperature range from 284.15 K to 353.15 K. A group additivity and molecular connectivity analysis: Legena

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https://en.wikipedia.org/wiki/Joback\_method

https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure

http://webbook.nist.gov/cgi/cbook.cgi?ID=C693583&Units=SI

https://www.chemeo.com/doc/models/crippen\_log10ws

https://www.cheric.org/research/kdb/hcprop/showprop.php?cmpid=1649 https://www.doi.org/10.1007/s10765-016-2064-y

cpg: Ideal gas heat capacity Liquid phase heat capacity cpl:

dvisc: Dynamic viscosity

gf: Standard Gibbs free energy of formation hf: Enthalpy of formation at standard conditions hfus: Enthalpy of fusion at standard conditions hfust: Enthalpy of fusion at a given temperature

Enthalpy of vaporization at standard conditions hvap: hvapt: Enthalpy of vaporization at a given temperature log10ws: Log10 of Water solubility in mol/llogp: Octanol/Water partition coefficientmcvol: McGowan's characteristic volume

pc: Critical Pressurepvap: Vapor pressure

rinpol: Non-polar retention indices

ripol: Polar retention indices speedsl: Speed of sound in fluid

**tb:** Normal Boiling Point Temperature

tc: Critical Temperature

tf: Normal melting (fusion) point

vc: Critical Volume

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