

# Eptam

## Other names:

Alirox  
Carbamic acid, dipropylthio-, S-ethyl ester  
Carbamothioic acid, N,N-dipropyl-, S-ethyl ester  
Carbamothioic acid, dipropyl-, S-ethyl ester  
Dipropylcarbamothioic acid S-ethyl ester  
EPTC  
EPTC (pesticide)  
EPTC (pesticide, MW 189)  
Eptam 6E  
Ethyl N,N-di-n-propylthiocarbamate  
Ethyl N,N-dipropylthiolcarbamate  
Ethyl di-n-propylthiolcarbamate  
FDA 1541  
Genep EPTC  
N,N-Dipropylthiocarbamic acid S-ethyl ester  
NSC 40486  
Niptan  
R-1608  
S-Aethyl-N,N-dipropylthiolcarbamat  
S-Ethyl Dipropylthiocarbamate  
S-Ethyl N,N-di-n-propylthiocarbamate  
S-Ethyl N,N-di-n-propylthiolcarbamate  
S-Ethyl N,N-dipropylthiocarbamate  
Stauffer R 1608  
Torbin  
Witox

**Inchi:** InChI=1S/C9H19NOS/c1-4-7-10(8-5-2)9(11)12-6-3/h4-8H2,1-3H3  
**InchiKey:** GUVLYNGULCJVDO-UHFFFAOYSA-N  
**Formula:** C9H19NOS  
**SMILES:** CCCN(CCC)C(=O)SCC  
**Mol. weight [g/mol]:** 189.32  
**CAS:** 759-94-4

## Physical Properties

Property code	Value	Unit	Source
gf	39.88	kJ/mol	Joback Method

hf	-232.27		kJ/mol	Joback Method
hfus	27.82		kJ/mol	Joback Method
hvap	51.23		kJ/mol	Joback Method
log10ws	-2.70			Aqueous Solubility Prediction Method
logp	2.981			Crippen Method
mcvol	165.570		ml/mol	McGowan Method
pc	2490.03		kPa	Joback Method
rinpol	1357.00			NIST Webbook
rinpol	1337.00			NIST Webbook
rinpol	1389.00			NIST Webbook
rinpol	1337.00			NIST Webbook
rinpol	1368.00			NIST Webbook
rinpol	1360.00			NIST Webbook
rinpol	1360.00			NIST Webbook
tb	540.41		K	Joback Method
tc	731.25		K	Joback Method
tf	307.99		K	Joback Method
vc	0.618		m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	387.44	J/mol×K	540.41	Joback Method
cpg	402.10	J/mol×K	572.22	Joback Method
cpg	416.05	J/mol×K	604.02	Joback Method
cpg	429.30	J/mol×K	635.83	Joback Method
cpg	441.87	J/mol×K	667.64	Joback Method
cpg	453.78	J/mol×K	699.44	Joback Method
cpg	465.05	J/mol×K	731.25	Joback Method

## Sources

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C759944&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**Aqueous Solubility Prediction Method:** <http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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