

N-phenyl-p-toluidine

Other names:	Benzenamine, 4-methyl-N-phenyl-
Inchi:	InChI=1S/C13H13N/c1-11-7-9-13(10-8-11)14-12-5-3-2-4-6-12/h2-10,14H,1H3
InchiKey:	AGHYMXKKEXDUTA-UHFFFAOYSA-N
Formula:	C13H13N
SMILES:	Cc1ccc(Nc2ccccc2)cc1
Mol. weight [g/mol]:	183.25
CAS:	620-84-8

Physical Properties

Property code	Value	Unit	Source
chs	-7026.60 ± 7.10	kJ/mol	NIST Webbook
gf	363.16	kJ/mol	Joback Method
hf	203.41	kJ/mol	Joback Method
hfs	48.90 ± 7.10	kJ/mol	NIST Webbook
hfus	22.22	kJ/mol	Joback Method
hvap	56.18	kJ/mol	Joback Method
log10ws	-3.75		Crippen Method
logp	3.739		Crippen Method
mcvol	156.490	ml/mol	McGowan Method
pc	3065.95	kPa	Joback Method
tb	605.35	K	Joback Method
tc	849.03	K	Joback Method
tf	354.29	K	Joback Method
vc	0.583	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	374.27	J/mol×K	605.35	Joback Method
cpg	390.48	J/mol×K	645.96	Joback Method
cpg	405.44	J/mol×K	686.58	Joback Method
cpg	419.22	J/mol×K	727.19	Joback Method
cpg	431.91	J/mol×K	767.80	Joback Method
cpg	443.56	J/mol×K	808.41	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C620848&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

chs:	Standard solid enthalpy of combustion
cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfs:	Solid phase enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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