

3-tert-Butyl-2-pyrazolin-5-one

Inchi:	InChI=1S/C7H12N2O/c1-7(2,3)5-4-6(10)9-8-5/h4H2,1-3H3,(H,9,10)
InchiKey:	VDNOHRWEOSDGQX-UHFFFAOYSA-N
Formula:	C7H12N2O
SMILES:	CC(C)(C)C1=NNC(=O)C1
Mol. weight [g/mol]:	140.18
CAS:	29211-68-5

Physical Properties

Property code	Value	Unit	Source
gf	157.39	kJ/mol	Joback Method
hf	-98.35	kJ/mol	Joback Method
hfus	14.41	kJ/mol	Joback Method
hvap	48.62	kJ/mol	Joback Method
log10ws	-1.52		Crippen Method
logp	0.908		Crippen Method
mcvol	115.860	ml/mol	McGowan Method
pc	4026.13	kPa	Joback Method
tb	550.49	K	Joback Method
tc	800.47	K	Joback Method
tf	444.28	K	Joback Method
vc	0.438	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	288.33	J/molxK	550.49	Joback Method
cpg	304.39	J/molxK	592.15	Joback Method
cpg	319.49	J/molxK	633.82	Joback Method
cpg	333.60	J/molxK	675.48	Joback Method
cpg	346.73	J/molxK	717.14	Joback Method
cpg	358.86	J/molxK	758.81	Joback Method
cpg	369.99	J/molxK	800.47	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C29211685&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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