

# 2,6-Dibromotoluene

<b>Other names:</b>	Benzene, 1,3-dibromo-2-methyl-
<b>Inchi:</b>	InChI=1S/C7H6Br2/c1-5-6(8)3-2-4-7(5)9/h2-4H,1H3
<b>InchiKey:</b>	OCSKCBIGEMSDIS-UHFFFAOYSA-N
<b>Formula:</b>	C7H6Br2
<b>SMILES:</b>	Cc1c(Br)cccc1Br
<b>Mol. weight [g/mol]:</b>	249.93
<b>CAS:</b>	69321-60-4

## Physical Properties

Property code	Value	Unit	Source
gf	129.85	kJ/mol	Joback Method
hf	78.44	kJ/mol	Joback Method
hfus	17.72	kJ/mol	Joback Method
hvap	47.65	kJ/mol	Joback Method
log10ws	-4.27		Crippen Method
logp	3.520		Crippen Method
mcvol	120.730	ml/mol	McGowan Method
pc	4749.69	kPa	Joback Method
tb	528.52	K	Joback Method
tc	781.02	K	Joback Method
tf	339.71	K	Joback Method
vc	0.444	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	203.12	J/mol×K	528.52	Joback Method
cpg	212.22	J/mol×K	570.60	Joback Method
cpg	220.62	J/mol×K	612.69	Joback Method
cpg	228.36	J/mol×K	654.77	Joback Method
cpg	235.51	J/mol×K	696.85	Joback Method
cpg	242.11	J/mol×K	738.94	Joback Method
cpg	248.21	J/mol×K	781.02	Joback Method
dvisc	0.0014758	Paxs	339.71	Joback Method

dvisc	0.0010226	Paxs	371.18	Joback Method
dvisc	0.0007504	Paxs	402.65	Joback Method
dvisc	0.0005759	Paxs	434.12	Joback Method
dvisc	0.0004581	Paxs	465.58	Joback Method
dvisc	0.0003751	Paxs	497.05	Joback Method
dvisc	0.0003145	Paxs	528.52	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C69321604&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C69321604&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcpvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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