

# 1-Naphthalenol, 2-nitroso-

<b>Other names:</b>	1-Naphthol, 2-nitroso- «beta»-Nitroso-«alpha»-naphthol «beta»-Nitrosonaphthol C.I. 76610 2-Nitroso-1-Naphthol 2-Nitrosonaphthol 1,2-naphthoquinone 2-oxime
<b>Inchi:</b>	InChI=1S/C10H7NO2/c12-10-8-4-2-1-3-7(8)5-6-9(10)11-13/h1-6,12H
<b>InchiKey:</b>	SYUYTOYKQOAVDW-UHFFFAOYSA-N
<b>Formula:</b>	C10H7NO2
<b>SMILES:</b>	O=Nc1ccc2ccccc2c1O
<b>Mol. weight [g/mol]:</b>	173.17
<b>CAS:</b>	132-53-6

## Physical Properties

Property code	Value	Unit	Source
chs	-4873.60 ± 2.30	kJ/mol	NIST Webbook
chs	-4884.40	kJ/mol	NIST Webbook
hf	-5.44	kJ/mol	NIST Webbook
hsub	56.50 ± 4.20	kJ/mol	NIST Webbook
hvap	64.54	kJ/mol	Joback Method
log10ws	-3.58		Crippen Method
logp	2.943		Crippen Method
mvol	125.960	ml/mol	McGowan Method
pc	4528.58	kPa	Joback Method
tb	622.86	K	Joback Method
tc	865.96	K	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C132536&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C132536&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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