

METHYL 2-ETHYLHEXYL PHTHALATE

Other names:	1,2-Benzenedicarboxylic acid, 2-ethylhexyl methyl ester 2-Ethylhexyl methyl phthalate
Inchi:	InChI=1S/C17H24O4/c1-4-6-9-13(5-2)12-21-17(19)15-11-8-7-10-14(15)16(18)20-3/h7-8,
InchiKey:	MQYNKBGLFNKJPL-UHFFFAOYSA-N
Formula:	C17H24O4
SMILES:	CCCCC(CC)COC(=O)c1cccc1C(=O)OC
Mol. weight [g/mol]:	292.37

Physical Properties

Property code	Value	Unit	Source
hf	-752.19	kJ/mol	Cheméo Relay (1.0)
hfus	35.49	kJ/mol	Joback Method
hvap	89.21	kJ/mol	Cheméo Relay (1.0)
ie	9.13	eV	Cheméo Relay (1.0)
log10ws	-4.88		Cheméo Relay (1.0)
logp	3.846		Crippen Method
mcvol	241.510	ml/mol	McGowan Method
pc	1687.95	kPa	Joback Method
rinpol	2053.00		NIST Webbook
rinpol	2053.00		NIST Webbook
tb	607.97	K	Cheméo Relay (1.0)
tc	808.22	K	Cheméo Relay (1.0)
tf	235.72	K	Cheméo Relay (1.0)

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	704.89	J/mol×K	772.16	Joback Method
cpg	783.29	J/mol×K	975.26	Joback Method
cpg	772.73	J/mol×K	941.41	Joback Method
cpg	761.19	J/mol×K	907.56	Joback Method
cpg	748.65	J/mol×K	873.71	Joback Method
cpg	735.10	J/mol×K	839.86	Joback Method
cpg	720.52	J/mol×K	806.01	Joback Method

dvisc	0.0001882	Pa×s	610.88	Joback Method
dvisc	0.0000756	Pa×s	772.16	Joback Method
dvisc	0.0000979	Pa×s	718.40	Joback Method
dvisc	0.0001322	Pa×s	664.64	Joback Method
dvisc	0.0009014	Pa×s	449.61	Joback Method
dvisc	0.0002868	Pa×s	557.13	Joback Method
dvisc	0.0004783	Pa×s	503.37	Joback Method

Sources

Crippen Method: https://www.chemeo.com/doc/models/crippen_log10ws

Cheméo Relay (1.0):

Cheméo Relay (1.0, beta):

NIST Webbook: <http://webbook.nist.gov/cgi/cbook.cgi?ID=C56166837&Units=SI>

Joback Method: https://en.wikipedia.org/wiki/Joback_method

McGowan Method: <http://link.springer.com/article/10.1007/BF02311772>

Crippen Method: <http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point

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