

# 2H-Thiopyran-3(4H)-one, dihydro-

<b>Other names:</b>	Dihydro-2H-thiopyran-3(4H)-one Tetrahydrothiopyran-3-one 1-Thiacyclohexan-3-one 3-Oxotetrahydrothiopyran 3-Thiacyclohexanone 5,6-Dihydro-2(H)-thiopyran-3(4H)-one Tetrahydrothia-3-pyranone
<b>Inchi:</b>	InChI=1S/C5H8OS/c6-5-2-1-3-7-4-5/h1-4H2
<b>InchiKey:</b>	ATAMXDLUUTYFKT-UHFFFAOYSA-N
<b>Formula:</b>	C5H8OS
<b>SMILES:</b>	O=C1CCCSC1
<b>Mol. weight [g/mol]:</b>	116.18
<b>CAS:</b>	19090-03-0

## Physical Properties

Property code	Value	Unit	Source
chl	-3531.00 ± 2.50	kJ/mol	NIST Webbook
gf	-59.35	kJ/mol	Joback Method
hf	-133.00 ± 2.60	kJ/mol	NIST Webbook
hfus	2.64	kJ/mol	Joback Method
hvap	49.30 ± 0.10	kJ/mol	NIST Webbook
log10ws	-0.97		Crippen Method
logp	1.083		Crippen Method
mcvol	88.370	ml/mol	McGowan Method
pc	4883.38	kPa	Joback Method
ripol	1541.00		NIST Webbook
tb	453.67	K	Joback Method
tc	701.30	K	Joback Method
tf	309.40	K	Joback Method
vc	0.302	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	165.39	J/mol×K	453.67	Joback Method
cpg	178.19	J/mol×K	494.94	Joback Method
cpg	190.36	J/mol×K	536.21	Joback Method
cpg	201.88	J/mol×K	577.49	Joback Method
cpg	212.76	J/mol×K	618.76	Joback Method
cpg	222.98	J/mol×K	660.03	Joback Method
cpg	232.53	J/mol×K	701.30	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C19090030&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C19090030&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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