

Ethyl perfluorohexanoate

Other names:	2,2,3,3,4,4,5,5,6,6,6-Undecafluoro- hexanoic acid ethyl ester
Inchi:	InChI=1S/C8H5F11O2/c1-2-21-3(20)4(9,10)5(11,12)6(13,14)7(15,16)8(17,18)19/h2H2,1
InchiKey:	PIGASTCSFHUJBM-UHFFFAOYSA-N
Formula:	C8H5F11O2
SMILES:	CCOC(=O)C(F)(F)C(F)(F)C(F)(F)C(F)(F)C(F)(F)C(F)(F)F
Mol. weight [g/mol]:	342.11

Physical Properties

Property code	Value	Unit	Source
gf	-2346.15	kJ/mol	Joback Method
hf	-2654.21	kJ/mol	Joback Method
hfus	16.07	kJ/mol	Joback Method
hvap	27.09	kJ/mol	Joback Method
log10ws	-3.95		Crippen Method
logp	3.653		Crippen Method
mvol	150.490	ml/mol	McGowan Method
pc	1826.28	kPa	Joback Method
rinpol	666.00		NIST Webbook
rinpol	666.40		NIST Webbook
tb	434.55	K	Joback Method
tc	572.05	K	Joback Method
tf	270.67	K	Joback Method
vc	0.650	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	381.88	J/mol×K	434.55	Joback Method
cpg	393.71	J/mol×K	457.47	Joback Method
cpg	404.79	J/mol×K	480.38	Joback Method
cpg	415.15	J/mol×K	503.30	Joback Method
cpg	424.83	J/mol×K	526.22	Joback Method
cpg	433.85	J/mol×K	549.13	Joback Method
cpg	442.24	J/mol×K	572.05	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R70094&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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