

# Citric acid, trimethyl ester

<b>Other names:</b>	1,2,3-Propanetricarboxylic acid, 2-hydroxy-, trimethyl ester Trimethyl citrate 3-Hydroxy-3-methoxycarbonylpentanedioic acid, dimethyl ester
<b>Inchi:</b>	InChI=1S/C9H14O7/c1-14-6(10)4-9(13,8(12)16-3)5-7(11)15-2/h13H,4-5H2,1-3H3
<b>InchiKey:</b>	HDDLZVZGOWKFW-UHFFFAOYSA-N
<b>Formula:</b>	C9H14O7
<b>SMILES:</b>	COC(=O)CC(O)(CC(=O)OC)C(=O)OC
<b>Mol. weight [g/mol]:</b>	234.20
<b>CAS:</b>	1587-20-8

## Physical Properties

Property code	Value	Unit	Source
gf	-810.84	kJ/mol	Joback Method
hf	-1124.47	kJ/mol	Joback Method
hfus	24.10	kJ/mol	Joback Method
hvap	78.48	kJ/mol	Joback Method
log10ws	0.45		Crippen Method
logp	-0.983		Crippen Method
mcvol	165.860	ml/mol	McGowan Method
pc	2976.29	kPa	Joback Method
rinpol	1442.00		NIST Webbook
rinpol	1442.00		NIST Webbook
rinpol	1424.00		NIST Webbook
rinpol	1442.00		NIST Webbook
rinpol	1461.00		NIST Webbook
rinpol	1461.00		NIST Webbook
rinpol	1424.00		NIST Webbook
rinpol	1442.00		NIST Webbook
tb	723.14	K	Joback Method
tc	912.95	K	Joback Method
tf	470.91	K	Joback Method
vc	0.620	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	461.78	J/molxK	723.14	Joback Method
cpg	471.58	J/molxK	754.77	Joback Method
cpg	480.77	J/molxK	786.41	Joback Method
cpg	489.33	J/molxK	818.04	Joback Method
cpg	497.27	J/molxK	849.68	Joback Method
cpg	504.58	J/molxK	881.31	Joback Method
cpg	511.25	J/molxK	912.95	Joback Method
dvisc	0.0006814	Paxs	470.91	Joback Method
dvisc	0.0003268	Paxs	512.95	Joback Method
dvisc	0.0001751	Paxs	554.99	Joback Method
dvisc	0.0001025	Paxs	597.02	Joback Method
dvisc	0.0000644	Paxs	639.06	Joback Method
dvisc	0.0000428	Paxs	681.10	Joback Method
dvisc	0.0000299	Paxs	723.14	Joback Method
hvapt	617.40	kJ/mol	469.50	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1587208&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1587208&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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