

# Decane, 1,10-dibromo-

<b>Other names:</b>	1,10-Dibromodecane DECAMETHYLENE DIBROMIDE NSC 6086 a,w-Dibromodecane
<b>Inchi:</b>	InChI=1S/C10H20Br2/c11-9-7-5-3-1-2-4-6-8-10-12/h1-10H2
<b>InchiKey:</b>	GTQHJCOHNAFHRE-UHFFFAOYSA-N
<b>Formula:</b>	C10H20Br2
<b>SMILES:</b>	BrCCCCCCCCCBr
<b>Mol. weight [g/mol]:</b>	300.07
<b>CAS:</b>	4101-68-2

## Physical Properties

Property code	Value	Unit	Source
gf	61.96	kJ/mol	Joback Method
hf	-197.07	kJ/mol	Joback Method
hfus	32.23	kJ/mol	Joback Method
hvap	50.72	kJ/mol	Joback Method
log10ws	-4.87		Crippen Method
logp	4.897		Crippen Method
mcvol	186.760	ml/mol	McGowan Method
pc	2402.92	kPa	Joback Method
rinpol	1735.00		NIST Webbook
tb	560.52	K	Joback Method
tc	751.80	K	Joback Method
tf	322.06	K	Joback Method
vc	0.720	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	410.32	J/molxK	560.52	Joback Method
cpg	473.89	J/molxK	719.92	Joback Method
cpg	462.46	J/molxK	688.04	Joback Method
cpg	450.43	J/molxK	656.16	Joback Method

cpg	437.75	J/molxK	624.28	Joback Method
cpg	424.39	J/molxK	592.40	Joback Method
cpg	484.74	J/molxK	751.80	Joback Method
dvisc	0.0002450	Paxs	560.52	Joback Method
dvisc	0.0003140	Paxs	520.78	Joback Method
dvisc	0.0004192	Paxs	481.03	Joback Method
dvisc	0.0005896	Paxs	441.29	Joback Method
dvisc	0.0008871	Paxs	401.55	Joback Method
dvisc	0.0014602	Paxs	361.80	Joback Method
dvisc	0.0027180	Paxs	322.06	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	433.20	K	2.00	NIST Webbook
tbrp	433.00	K	2.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.67728e+01
Coeff. B	-5.46929e+03
Coeff. C	-9.71500e+01
Temperature range (K), min.	428.92
Temperature range (K), max.	574.35

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.thermo.com/files/research/kdb/mol/mol1651.mol">https://www.thermo.com/files/research/kdb/mol/mol1651.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4101682&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4101682&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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