

# Diethylmalonic acid, ethyl 2,2,2-trichloroethyl ester

Inchi:	InChI=1S/C11H17Cl3O4/c1-4-10(5-2,8(15)17-6-3)9(16)18-7-11(12,13)14/h4-7H2,1-3H3
InchiKey:	ICDBZPXXABQWIL-UHFFFAOYSA-N
Formula:	C11H17Cl3O4
SMILES:	CCOC(=O)C(CC)(CC)C(=O)OCC(Cl)(Cl)Cl
Mol. weight [g/mol]:	319.61

## Physical Properties

Property code	Value	Unit	Source
gf	-456.21	kJ/mol	Joback Method
hf	-824.69	kJ/mol	Joback Method
hfus	27.58	kJ/mol	Joback Method
hvap	68.95	kJ/mol	Joback Method
log10ws	-3.47		Crippen Method
logp	3.269		Crippen Method
mcvol	217.450	ml/mol	McGowan Method
pc	1966.56	kPa	Joback Method
rinpol	1602.00		NIST Webbook
tb	709.49	K	Joback Method
tc	918.24	K	Joback Method
tf	452.65	K	Joback Method
vc	0.825	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	551.78	J/molxK	709.49	Joback Method
cpg	563.95	J/molxK	744.28	Joback Method
cpg	575.26	J/molxK	779.07	Joback Method
cpg	585.75	J/molxK	813.86	Joback Method
cpg	595.46	J/molxK	848.66	Joback Method
cpg	604.42	J/molxK	883.45	Joback Method
cpg	612.69	J/molxK	918.24	Joback Method
dvisc	0.0010196	Paxs	452.65	Joback Method
dvisc	0.0005666	Paxs	495.46	Joback Method

dvisc	0.0003457	Paxs	538.26	Joback Method
dvisc	0.0002269	Paxs	581.07	Joback Method
dvisc	0.0001577	Paxs	623.88	Joback Method
dvisc	0.0001149	Paxs	666.68	Joback Method
dvisc	0.0000870	Paxs	709.49	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370470&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370470&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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