

# Bamethan

**Other names:**

Benzenemethanol, «alpha»-[(butylamino)methyl]-4-hydroxy-  
Benzyl alcohol, «alpha»-((butylamino)methyl)-p-hydroxy-  
Bamethane  
Butedrine  
«alpha»-((Butylamino)methyl)-4-hydroxybenzenemethanol  
2-Butylamino-1-p-hydroxyphenylethanol  
«alpha»-((Butylamino)methyl)-p-hydroxybenzyl alcohol  
N-Dutylnorsympathol  
Butyl-nor-sympatol  
N-Butylnorsynephrine  
Butylsympathol  
1-(p-Hydroxyphenyl)-2-butylaminoethanol  
1-(4-Hydroxyphenyl)-1-hydroxy-2-butylaminoethane  
«alpha»-[(Butylamino)methyl]-p-hydroxybenzyl, «alpha»-hydroxyl-  
(./-.)-Bamethane  
dl-Bamethane

**Inchi:**

InChI=1S/C12H19NO2/c1-2-3-8-13-9-12(15)10-4-6-11(14)7-5-10/h4-7,12-15H,2-3,8-9H2

**InchiKey:**

RDUHXGIIUDVSHR-UHFFFAOYSA-N

**Formula:**

C12H19NO2

**SMILES:**

CCCCNCC(O)c1ccc(O)cc1

**Mol. weight [g/mol]:**

209.28

**CAS:**

3703-79-5

## Physical Properties

Property code	Value	Unit	Source
gf	-41.92	kJ/mol	Joback Method
hf	-335.83	kJ/mol	Joback Method
hfus	32.32	kJ/mol	Joback Method
hvap	80.32	kJ/mol	Joback Method
log10ws	-2.41		Crippen Method
logp	1.815		Crippen Method
mcvol	177.900	ml/mol	McGowan Method
pc	3166.83	kPa	Joback Method
rinpola	1735.00		NIST Webbook
tb	723.17	K	Joback Method
tc	925.14	K	Joback Method
tf	461.62	K	Joback Method

vc

0.614

m<sup>3</sup>/kmol

Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	512.35	J/mol×K	723.17	Joback Method
cpg	524.70	J/mol×K	756.83	Joback Method
cpg	536.38	J/mol×K	790.49	Joback Method
cpg	547.46	J/mol×K	824.15	Joback Method
cpg	558.01	J/mol×K	857.82	Joback Method
cpg	568.11	J/mol×K	891.48	Joback Method
cpg	577.83	J/mol×K	925.14	Joback Method

## Sources

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C3703795&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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