

# Acetaldehyde, phenylhydrazone

**Inchi:** InChI=1S/C8H10N2/c1-2-9-10-8-6-4-3-5-7-8/h2-7,10H,1H3  
**InchiKey:** KURBTRNHGDQKOS-UHFFFAOYSA-N  
**Formula:** C8H10N2  
**SMILES:** CC=NNc1ccccc1  
**Mol. weight [g/mol]:** 134.18  
**CAS:** 935-07-9

## Physical Properties

Property code	Value	Unit	Source
chs	-4436.30	kJ/mol	NIST Webbook
hf	163.77	kJ/mol	Joback Method
hfs	-141.00	kJ/mol	NIST Webbook
hfs	-167.00	kJ/mol	NIST Webbook
hvap	45.43	kJ/mol	Joback Method
log10ws	-2.05		Crippen Method
logp	2.104		Crippen Method
mcvol	115.480	ml/mol	McGowan Method
pc	3276.53	kPa	Joback Method
tb	535.97	K	Joback Method
tc	768.26	K	Joback Method

## Sources

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)  
**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>  
**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C935079&Units=SI>  
**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>  
**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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