

# Diethylmalonic acid, heptyl 5-methoxy-3-methylpentyl ester

<b>Inchi:</b>	InChI=1S/C21H40O5/c1-6-9-10-11-12-15-25-19(22)21(7-2,8-3)20(23)26-17-14-18(4)13-1
<b>InchiKey:</b>	UJNURMSCZMKUMR-UHFFFAOYSA-N
<b>Formula:</b>	C21H40O5
<b>SMILES:</b>	CCCCCCCOC(=O)C(CC)(CC)C(=O)OCCC(C)CCOC
<b>Mol. weight [g/mol]:</b>	372.54

## Physical Properties

Property code	Value	Unit	Source
gf	-446.50	kJ/mol	Joback Method
hf	-1112.62	kJ/mol	Joback Method
hfus	45.97	kJ/mol	Joback Method
hvap	81.38	kJ/mol	Joback Method
log10ws	-4.94		Crippen Method
logp	4.912		Crippen Method
mcvol	327.500	ml/mol	McGowan Method
pc	1019.43	kPa	Joback Method
rinpol	2206.00		NIST Webbook
tb	851.21	K	Joback Method
tc	1043.75	K	Joback Method
tf	480.40	K	Joback Method
vc	1.260	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1063.19	J/molxK	851.21	Joback Method
cpg	1081.54	J/molxK	883.30	Joback Method
cpg	1098.68	J/molxK	915.39	Joback Method
cpg	1114.64	J/molxK	947.48	Joback Method
cpg	1129.44	J/molxK	979.57	Joback Method
cpg	1143.12	J/molxK	1011.66	Joback Method
cpg	1155.70	J/molxK	1043.75	Joback Method
dvisc	0.0005533	Paxs	480.40	Joback Method
dvisc	0.0002455	Paxs	542.20	Joback Method

dvisc	0.0001286	Paxs	604.00	Joback Method
dvisc	0.0000760	Paxs	665.81	Joback Method
dvisc	0.0000491	Paxs	727.61	Joback Method
dvisc	0.0000340	Paxs	789.41	Joback Method
dvisc	0.0000248	Paxs	851.21	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370767&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370767&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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