

# Methanesulfonic acid

<b>Other names:</b>	Methanesulphonic acid Methylsulfonic acid Kyselina methansulfonova <chem>CH3SO3H</chem> NSC 3718
<b>Inchi:</b>	<chem>InChI=1S/CH4O3S/c1-5(2,3)4/h1H3,(H,2,3,4)</chem>
<b>InchiKey:</b>	AFVFQIVMOAPDHO-UHFFFAOYSA-N
<b>Formula:</b>	<chem>CH4O3S</chem>
<b>SMILES:</b>	<chem>CS(=O)(=O)O</chem>
<b>Mol. weight [g/mol]:</b>	96.11
<b>CAS:</b>	75-75-2

## Physical Properties

Property code	Value	Unit	Source
affp	761.30	kJ/mol	NIST Webbook
basg	728.90	kJ/mol	NIST Webbook
gf	-647.82	kJ/mol	Joback Method
hf	-669.55	kJ/mol	Joback Method
hfus	13.81	kJ/mol	Joback Method
hvap	53.13	kJ/mol	Joback Method
log10ws	0.61		Crippen Method
logp	-0.496		Crippen Method
mcvol	58.910	ml/mol	McGowan Method
pc	8783.57	kPa	Joback Method
tb	362.24	K	Joback Method
tc	523.75	K	Joback Method
tf	200.41	K	Joback Method
vc	0.236	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	98.89	J/mol×K	362.24	Joback Method
cpg	102.86	J/mol×K	389.16	Joback Method

cpg	106.75	J/mol×K	416.08	Joback Method
cpg	110.55	J/mol×K	442.99	Joback Method
cpg	114.26	J/mol×K	469.91	Joback Method
cpg	117.86	J/mol×K	496.83	Joback Method
cpg	121.36	J/mol×K	523.75	Joback Method
hvapt	73.90	kJ/mol	417.50	NIST Webbook

## Sources

Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C75752&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C75752&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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